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(54) ANTIMICROBIAL PEPTIDES AND METHODS OF USE

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ABSTRACT

Disclosed herein are novel antimicrobial peptides with useful, improved, or superior properties such as antimicrobial activity, specificity, resistance to degradation desirable levels of hemolytic activity, and therapeutic index against a broad range of microorganisms including gram-negative and grampositive bacteria and other organisms having a cellular or structural component of a lipid bilayer membrane. Also provided are methods of making and using such peptides to control microbial growth and in pharmaceutical compositions for treatment or prevention of infections caused by such microorganisms. Certain peptides result from structure-based rational design relating to antimicrobial peptide V681, with single D-/L-amino acid substitutions or charged residue substitutions in or near the center of the peptide on the nonpolar or polar face. Some peptides contain one or more (or all) amino acids in the D configuration. Compositions disclosed herein have useful clinical potential as antibiotics including broad spectrum antibiotics.

10 Claims, 15 Drawing Sheets

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* cited by examiner

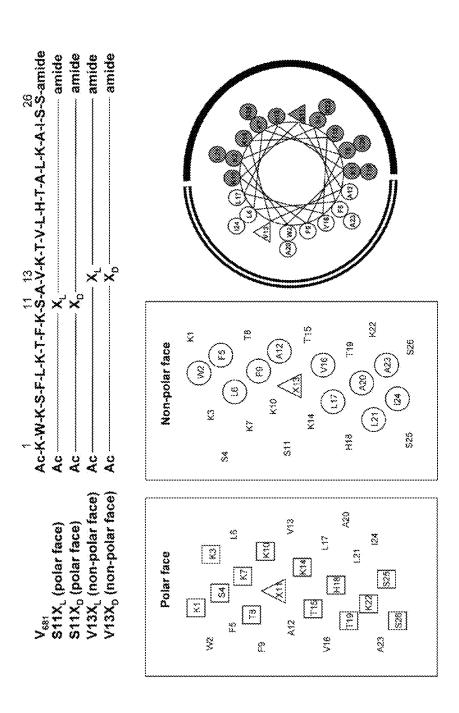


Figure 1

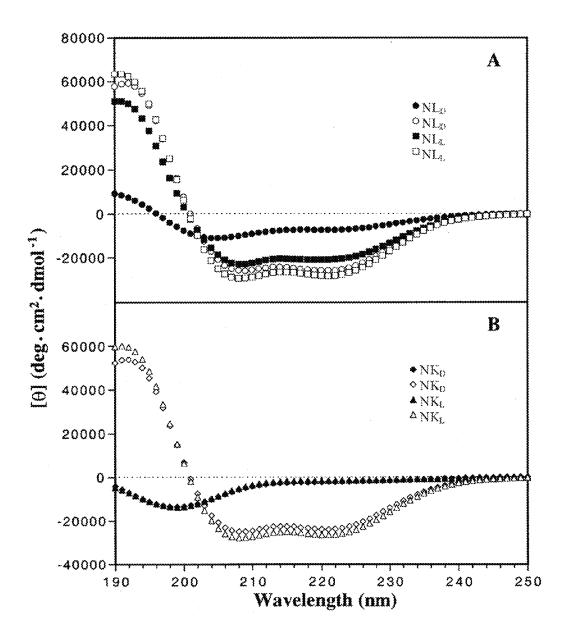


Figure 2

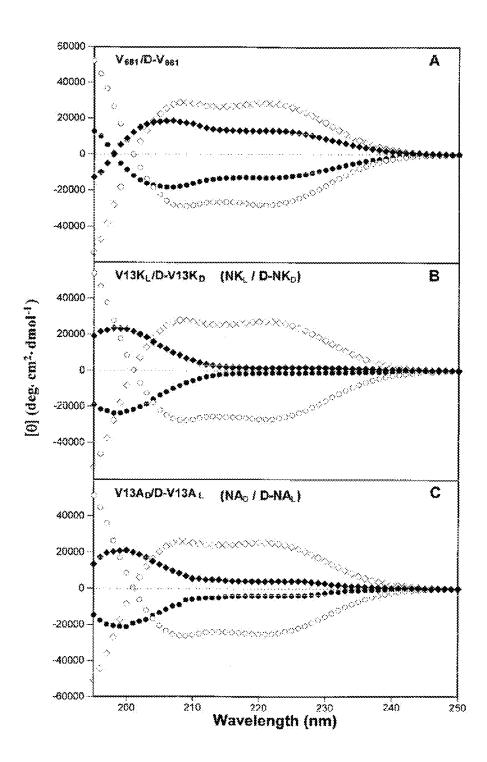


Figure 3

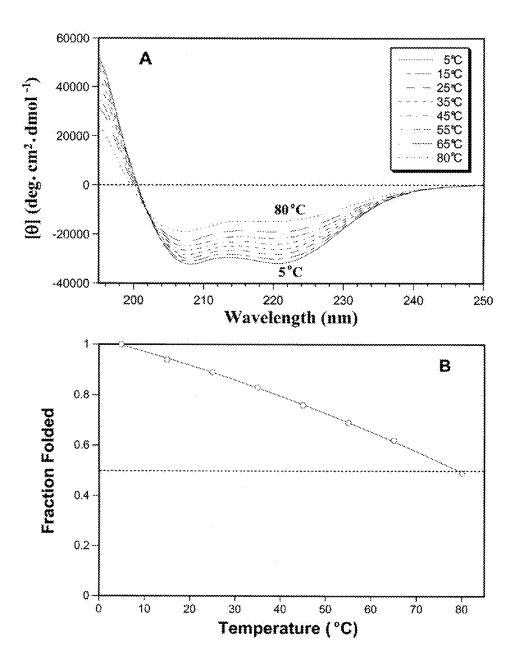


Figure 4

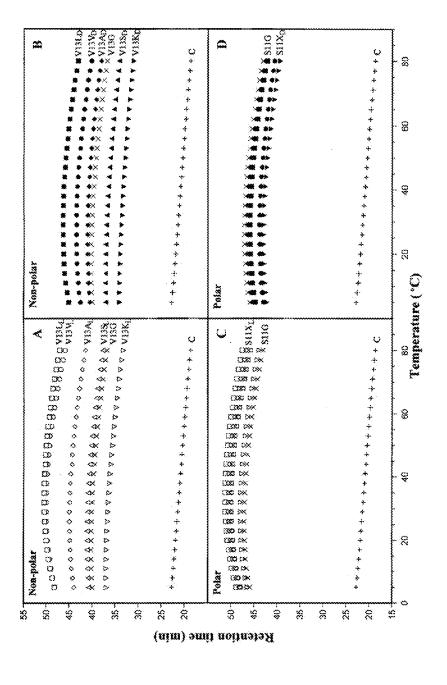


Figure (

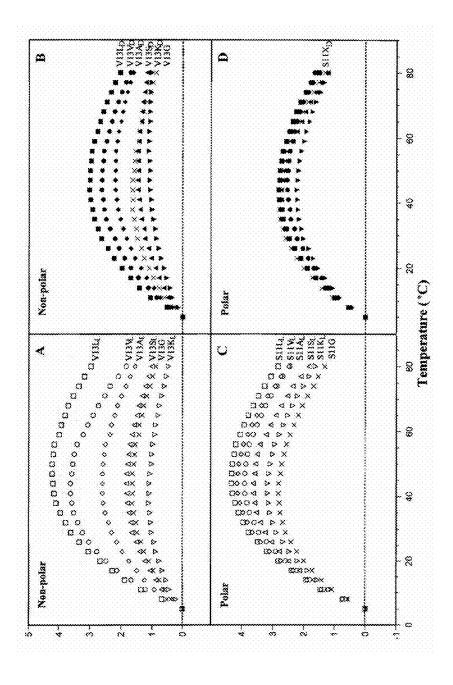


Figure 6

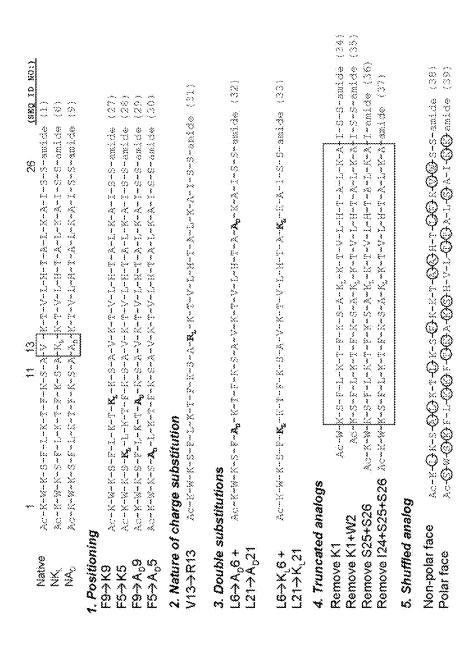


Figure 7A

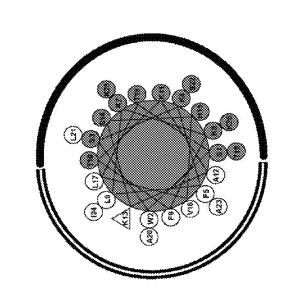
Figure 7B

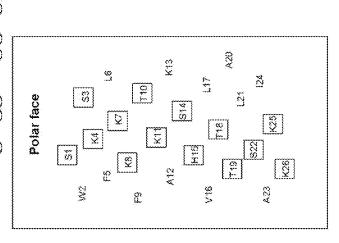
AC-K(BK-S(AXBK-T(BK-S(B)K-K-T(AXB)H-T(LXB)K(YM)S-S-amide Ac-K-W-K-S-F-L-K-T-F-K-S-A-K-K-T-V-L-H-T-A-L-K-A-I-S-S-amide ž Non-polar face 2

May 24, 2016

<u>7</u> Ş S 38 NK, Shuffled Analog







Ac-K-W-K-S-F-L-K-T-F-K-S-A-K-Y-T-V-L-H-T-A-L-K-A-I-S-S-amide ROGS-WGSFB-1-(FKF) FFFB-A-KSFH-V-1-(FFB-1-(S)A-1-FFF) amide 1 2 3 4 5 6 7 8 91011121314151617181920212223242526 NK, Shuffled Analog

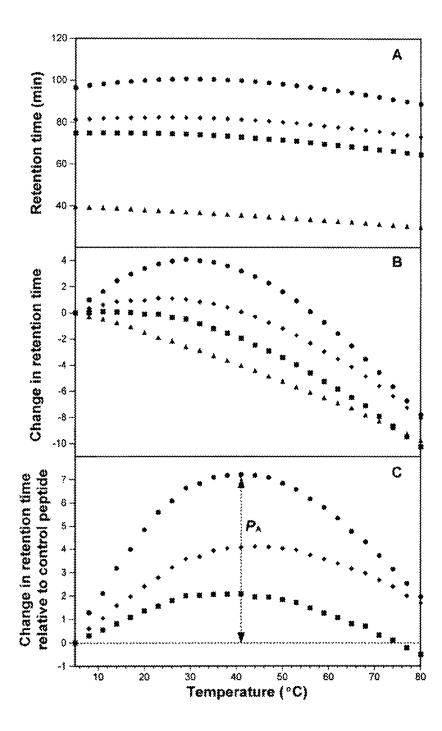


Figure 8

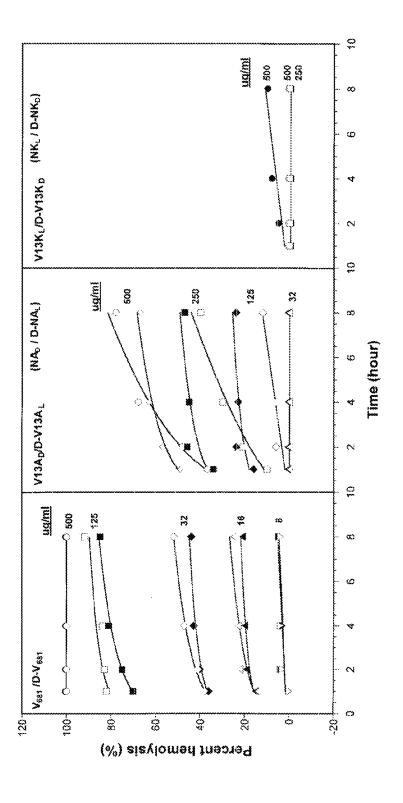


Figure 9

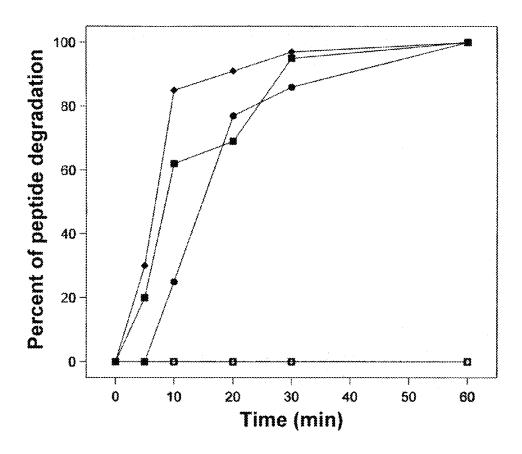
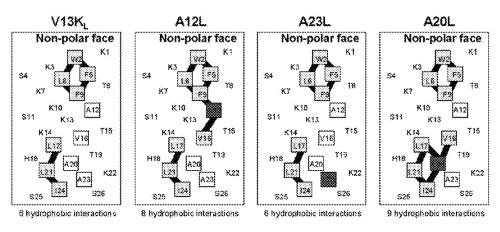
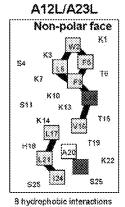
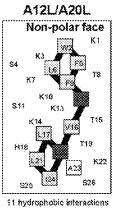
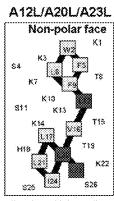


Figure 10









. Trily or opinions offersomer

12 hydrophobic interactions

Figure 11

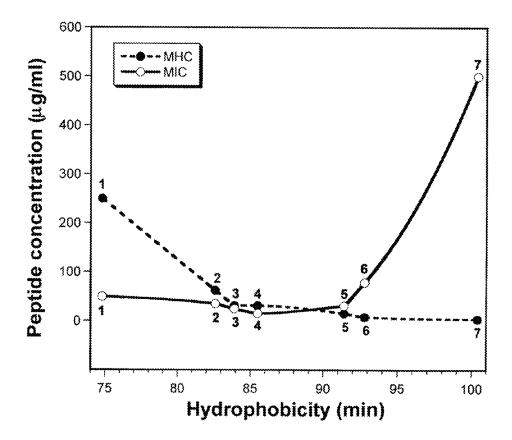


Figure 12

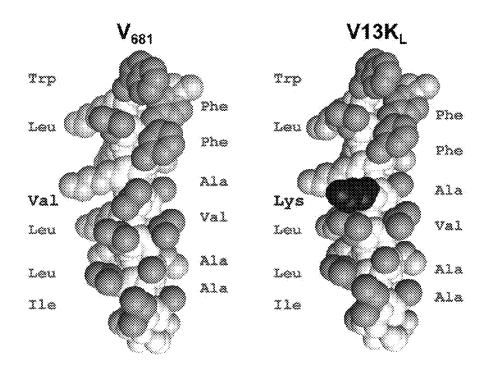


Figure 13

ANTIMICROBIAL PEPTIDES AND METHODS OF USE

CROSS-REFERENCE TO RELATED APPLICATIONS

This application is a Continuation of U.S. application Ser. No. 11/721,915, filed Jun. 6, 2008, now U.S. Pat. No. 8,252, 737, which is the U.S. National Stage of International Application No. PCT/US2005/045393, filed Dec. 15, 2005, which ¹⁰ claims benefit of U.S. Provisional Application No. 60/636, 220, filed Dec. 15, 2004, all of which are hereby incorporated by reference to the extent not inconsistent with the present disclosure.

STATEMENT ON FEDERALLY SPONSORED RESEARCH OR DEVELOPMENT

This invention was made with government support under contract number AI048717 awarded by the National Institutes of Health. The government has certain rights in the invention.

BACKGROUND OF THE INVENTION

The present invention broadly relates to novel antimicrobial peptides and methods of making and using such peptides to inhibit microbial growth and in pharmaceutical compositions for treatment or prevention of infections caused by a broad range of microorganisms including gram-positive and 30 gram-negative bacteria.

The extensive clinical use of classical antibiotics has led to the growing emergence of many medically relevant resistant strains of bacteria (1,2). Moreover, only three new structural classes of antibiotics (the oxazolidinone, linezolid, the strep- 35 togramins and the lipopeptide-daptomycin) have been introduced into medical practice in the past 40 years. Therefore, the development of a new class of antibiotics has great significance. The cationic antimicrobial peptides could represent such a new class of antibiotics (3-5). Although the exact 40 mode of action of the cationic antimicrobial peptides has not been established, all cationic amphipathic peptides interact with membranes and it has been proposed that the cytoplasmic membrane is the main target of some peptides, where peptide accumulation in the membrane may cause increased 45 permeability and loss of barrier function (6,7). Therefore, the development of resistance to these membrane active peptides is less likely because this would require substantial changes in the lipid composition of cell membranes of microorganisms.

Two major classes of the cationic antimicrobial peptides 50 are the α -helical and the β -sheet peptides (3,4,8,9). The β -sheet class includes cyclic peptides constrained in this conformation either by intramolecular disulfide bonds, e.g., defensins (10) and protegrins (11), or by an N-terminal to C-terminal covalent bond, e.g., gramicidin S (12) and tyrocidines (13). Unlike the β -sheet peptides, α -helical peptides are more linear molecules that mainly exist as disordered structures in aqueous media and become amphipathic helices upon interaction with the hydrophobic membranes, e.g., cecropins (14), magainins (15) and melittins (16). Concerning such peptides, we have explored certain factors that may be important in relation to antimicrobial activity.

The major barrier to the use of antimicrobial peptides as antibiotics is their toxicity or ability to lyse eukaryotic cells. This is perhaps not a surprising result if the target is indeed the 65 cell membrane (3-6). To be useful as a broad-spectrum antibiotic, it is necessary to dissociate anti-eukaryotic activity

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from antimicrobial activity, i.e., increasing the antimicrobial activity and reducing toxicity to normal cells.

A synthetic peptide approach to examining the effect of changes, including small or incremental changes, in hydrophobicity/hydrophilicity, amphipathicity and helicity of cationic antimicrobial peptides can facilitate rapid progress in rational design of peptide antibiotics. Generally, only L-amino acids are the isomers found throughout natural peptides and proteins; D-amino acids are the isomeric forms rarely seen in natural peptides/proteins except in some bacterial cell walls. In certain circumstances, the helix-destabilizing properties of D-amino acids offer a potential systematic approach to the controlled alteration of the hydrophobicity, amphipathicity, and helicity of amphipathic 15 α-helical model peptides (26).

Herein, we disclose the utilization of the structural framework of an amphipathic α -helical antimicrobial peptide, V_{681} (28), to systematically change peptide amphipathicity, hydrophobicity and helicity by single D- or L-amino acid substitutions in the center of either the polar or nonpolar faces of the amphipathic helix. Peptide V_{681} has been shown to have excellent antimicrobial activity and strong hemolytic activity (27,28) and thus served as a potentially useful candidate for our study. By introducing different D- or L-amino acid substitutions, we report here that hydrophobicity/amphipathicity and helicity have dramatic effects on the biophysical and biological activities and, utilizing this method, a significant improvement in antimicrobial activity and specificity can be achieved. In addition, high peptide hydrophobicity and amphipathicity can also result in greater peptide self-association in solution. Since we have developed a useful method to measure self-association of small amphipathic molecules, namely temperature profiling in reversed-phase chromatography (29,30), this technique was applied for the first time to investigate the influence of peptide dimerization ability on biological activities of α -helical antimicrobial peptides. The advantage of the invention will become apparent in the following description.

SUMMARY OF THE INVENTION

The present invention provides peptide compounds useful as antimicrobial agents and related methods. In embodiments of the invention, the antimicrobial peptides range in size from about 23 to about 26 amino acids in length joined by peptide bonds and have a core amino acid sequence of about 21 amino acids. The amino acids in the peptide compounds can be all in the L configuration, all in the D configuration or in a combination of D or L configurations. The peptides have potent antimicrobial activities and are useful against bacteria, fungi, viruses, and protozoa. The peptides are generally effective in the context of any organism having a cellular or structural component of a lipid bilayer membrane. These peptides are effective compounds for use in human and/or veterinary medicine, or as agents in agricultural, food science, or industrial applications.

Without wishing to be bound by a particular theory, from numerous structure/activity studies on both natural and synthetic antimicrobial peptides it is believed that a number of factors are important for antimicrobial activity. These are identified as including the presence of both hydrophobic and basic residues, an amphipathic nature that segregates basic and hydrophobic residues, and an inducible or preformed secondary structure (α -helical or β -sheet). Also without wishing to be bound by a particular theory, it is believed that by substituting different D-amino acids into the center of the hydrophobic face of an amphipathic α -helical model peptide,

0 0 7,0 10,000 =

disruption of alpha-helical structure can occur. Different D-amino acids can disrupt α -helical structure to different degrees, whilst the destabilized structure can still be induced to fold into an α -helix in hydrophobic medium. An advantage of this method of single D- or L-amino acid substitutions at a specific site is that it provides an opportunity for greater understanding of the mechanism of action of these peptides.

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It is recognized that regardless of the ultimate correctness of any mechanistic explanation or hypothesis, the compositions and methods of the invention can nonetheless be operative and useful.

For certain α -helical and β -sheet peptides, there have been attempts to delineate features responsible for anti-eukaryotic or toxic activities and for antimicrobial activities. High amphipathicity (17-20), high hydrophobicity (17, 20-22), as well as high helicity or β -sheet structure (20,23,24) may correlate with increased toxicity as measured by hemolytic activity. In contrast, antimicrobial activity may be less dependent on these factors, compared with hemolytic activity (17-21, 23-25). Here, specificity (or therapeutic index (TI) which is defined as the ratio of hemolytic activity and antimicrobial activity) for bacteria over erythrocytes could be increased in one of three ways: increasing antimicrobial activity, decreasing hemolytic activity while maintaining antimicrobial activity, or a combination of both, increasing antimicrobial activity and decreasing hemolytic activity.

The invention provides a method of treating a patient in need of therapy comprising administering to the patient a peptide compound of the invention. The invention provides a method of treating a microbial infection. In a particular embodiment, the microbial infection involves one or more of a bacterium, a virus, a fungus, or a protozoan. In a particular embodiment, the microbial infection involves one or more of each of those kinds of organisms, e.g. two different kinds of bacteria, and so forth.

In an embodiment, the invention provides a method for increasing antimicrobial activity of a peptide compound. In an embodiment, the invention provides a method for decreas-

ing hemolytic activity of a peptide compound while maintaining antimicrobial activity or minimizing a reduction of antimicrobial activity. In an embodiment, the invention provides a method of increasing antimicrobial activity and decreasing hemolytic activity of a peptide compound while maintaining antimicrobial activity or minimizing a reduction of antimicrobial activity.

The inventors discovered the antimicrobial peptides disclosed herein based on the premise that controlled alteration of the hydrophobicity/hydrophilicity, amphipathicity and helicity of an alpha helical peptide would yield a peptide with useful or improved biological activity and specificity (e.g. improved therapeutic index). Exemplified herein are peptides derived from the 26-residue peptide sequence, Ac-KWKS-FLKTFKS-AVKTVLHTALKAISS-amide (V₆₈₁, SEQ ID NO: 1). The terms "derived from" or "derivative" are meant to indicate that the inventive peptides are the same or shorter than the V_{681} peptide in size and have one or more amino acid residues substituted, or a combination of both; further variations are also described herein. The peptide compound V_{681} was used as the framework to study the effects of peptide hydrophobicity/hydrophilicity, amphipathicity and helicity on biological activities, for example antimicrobial and hemolytic activities, by substituting one or more amino acid residues at certain locations. These locations can include points at or near the center of the polar and nonpolar faces of the amphipathic helix in addition to other locations. The peptide V_{681} is disclosed in Zhang et al., 1999 and Zhang et

In an embodiment, the invention provides compositions and methods relating to a peptide selected from the group consisting of SEQ ID NOS:2, 4-14, and 16-25 and other peptides as disclosed herein, e.g. in the Examples. It is noted that SEQ ID NO:1, peptide V681, is exactly equivalent to SEQ ID NOS:3 and 15. See Table 1 which includes peptides with substitutions on the nonpolar face at position X=13 and on the polar face at position X=11. See also Table 2 which includes other peptide analogs.

TABLE 1

							S	umm	ary c	f par	tial se	equen	ce lis	sting i	infor	matic	n.										
SEQ ID	Peptide											A	mino	Acid	l Pos	ition											
NO:	Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26
Enanti	iomer*	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
1	V681	K	W	K	S	F	L	K	T	F	K	S	Α	\mathbf{V}	K	Τ	V	L	Η	T	\mathbf{A}	L	K	Α	I	S	S
2	NL_L	K	W	K	S	F	L	K	T	F	K	S	Α	\mathbf{L}_L	K	T	V	L	Η	T	Α	L	K	A	Ι	S	S
3	NV_L	K	W	K	S	F	L	K	T	F	K	S	Α	\mathbf{V}_L	K	Τ	V	L	Η	T	Α	L	K	Α	I	\mathbf{S}	\mathbf{S}
4	NA_L	K	W	K	S	F	L	K	T	F	K	S	Α	\mathbf{A}_{L}	K	Τ	V	L	Η	T	Α	L	K	Α	I	\mathbf{S}	\mathbf{S}
5	NS_L	K	W	K	\mathbf{S}	F	L	K	T	F	K	S	Α	\mathbf{S}_{L}	K	Τ	\mathbf{V}	L	Η	T	Α	L	K	Α	Ι	S	\mathbf{S}
6	NK_L	K	W	K	S	F	L	K	Τ	F	K	S	A	\mathbf{K}_{L}	K	T	\mathbf{V}	L	Η	T	Α	L	K	Α	Ι	S	S
7	NL_D	K	W	K	S	F	L	K	T	F	K	S	Α	\mathbf{L}_D	K	Τ	V	L	Η	T	Α	L	K	Α	Ι	S	S
8	NV_D	K	W	K	S	F	L	K	Τ	F	K	S	Α	\mathbf{V}_D	K	Τ	V	L	Η	T	Α	L	K	Α	Ι	\mathbf{S}	\mathbf{S}
9	NA_D	K	W	K	S	F	L	K	T	F	K	S	Α	\mathbf{A}_{D}	K	T	\mathbf{V}	L	Η	T	Α	L	K	Α	Ι	S	S
10	NS_D	K	W	K	S	F	L	K	T	F	K	S	Α	$\mathbf{S}_{\!D}$	K	T	\mathbf{V}	L	Η	T	Α	L	K	Α	Ι	S	S
11	NK_D	K	W	K	\mathbf{S}	F	L	K	T	F	K	S	Α	\mathbf{K}_{D}		T	V	L	Η	T	Α	L	K	Α	I	\mathbf{S}	\mathbf{S}
12	NG	K	W	K	\mathbf{S}	F	L	K	T	F	K	S	Α	\mathbf{G}	K	Τ	V	L	Η	T	Α	L	K	Α	Ι	S	S
13	PL_L	K	W	K	\mathbf{S}	F	L	K	T	F	K	\mathbf{L}_{L}	Α	V	K	Τ	V	L	Η	T	Α	L	K	Α	Ι	S	\mathbf{S}
14	PA_L	K	W	K	S	F	L	K	Τ	F	K	\mathbf{A}_{L}	Α	V	K	Τ	V	L	Η	Τ	Α	L	K	Α	Ι	S	\mathbf{S}
15	PS_L	K	W	K	S	F	L	K	T	F	K	\mathbf{S}_L	A	V	K	T	V	L	Η	T	Α	L	K	A	Ι	S	S
16	PV_L	K	W	K	\mathbf{S}	F	L	K	T	F	K	\mathbf{V}_L	Α	V	K	Τ	V	L	Η	T	Α	L	K	Α	Ι	S	S
17	PK_L	K	W	K	S	F	L	K	T	F	K			V	K	Т	V	L	Η	T	Α	L	K	Α	Ι	S	S
18	PL_D	K	W	K	S	F	L	K	Τ	F	K	$\mathbf{L}_{\!D}$	\mathbf{A}	V	K	T	V	L	Η	T	Α	L	K	Α	Ι	S	S
19	PA_D	K	W	K	S	F	L	K	T	F	K	\mathbf{A}_{D}		V	K	Т	V	L	Η	T	Α	L	K	Α	Ι	S	S
20	PS_D	K	W	K	S	F	L	K	T	F	K	$\mathbf{S}_{\!D}$	A	V	K	Τ	\mathbf{V}	L	Η	T	Α	L	K	Α	Ι	S	\mathbf{S}
21	PV_D	K	W	K	S	F	L	K	Т	F	K	\mathbf{V}_D		V	K	Т	V	L	Η	T	Α	L	K	Α	Ι	S	S
22	PK_D	K	W	K	S	F	L	K	Т	F	K	$\mathbf{K}_{\!D}$	Α	V	K	Т	V	L	Η	T	Α	L	K	Α	Ι	S	S
23	PG	K	W	K	S	F	L	K	T	F	K	\mathbf{G}	Α	V	K	T	V	L	Η	T	Α	L	K	Α	Ι	S	S

TABLE 1-continued

	Summary of partial sequence listing information.																										
SEQ ID	Peptide											Д	mino	Acid	l Pos	ition											
NO:	Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26
Enant	iomer	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D	D
24	$D-NK_D$	K	W	K	\mathbf{S}	F	L	K	T	F	K	S	\mathbf{A}	K	K	T	V	L	Η	Τ	A	L	K	\mathbf{A}	I	\mathbf{S}	S
25	$D-NA_L$	K	W	K	S	F	L	K	Τ	F	K	S	\mathbf{A}	\mathbf{A}_{L}	K	T	V	L	Η	Τ	A	L	K	A	I	S	S
1	D-V681	K	W	K	S	F	L	K	T	F	K	\mathbf{S}	A	\mathbf{v}^{-}	K	Τ	V	L	Η	Τ	A	L	K	Α	I	\mathbf{S}	\mathbf{S}

^{*}L-enantiomer unless otherwise indicated in the Enantiomer column or subscript.

TABLE 2

							Sı	ımm	arv o	f part	ial se	eaner	ice lis	ting	infor	matic	on.										
SEQ ID	Peptide									2 10000		40.00				ositi											
NO:	Name	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	20
	Enantiomer*	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L	L
1	V681	K	W	K	S	F	L	K	T	F	K	\mathbf{S}	\mathbf{A}	\mathbf{V}	K	Τ	V	L	Η	T	A	L	K	A	I	S	S
27	F9 to K9	K	W	K	S	F	L	K	T	K	K	S	Α	\mathbf{V}	K	T	V	L	Η	T	A	L	K	A	I	S	S
28	F5 to K5	K	W	K	S	K	L	K	Τ	F	K	\mathbf{S}	A	\mathbf{V}	K	Τ	V	L	Η	T	A	L	K	A	I	S	S
29	F9 to A_D 9	K	W	K	S	F	L	K	T	\mathbf{A}_{D}	K	S	A	\mathbf{V}	K	T	V	L	Η	T	A	L	K	A	I	S	S
30	F5 to A_D 5	K	W	K	S	\mathbf{A}_{D}	L	K	T	F	K	S	Α	\mathbf{V}	K	T	V	L	Η	T	A	L	K	A	I	S	S
31	V13 to R13	K	W	K	S	F	L	K	T	F	K	S	A	R	K	T	V	L	Η	T	A	L	K	A	Ι	S	S
32	$\text{L6-A}_D 6,$	K	W	K	S	F	$\mathbf{A}_{\!D}$	K	Τ	F	K	\mathbf{S}	A	\mathbf{V}	K	Τ	V	L	Η	T	A	\mathbf{A}_{D}	K	A	Ι	\mathbf{S}	S
	$L21-A_D21$																										
33	$L6-K_L6$, $L21-K_L21$	K	W	K	S	F	K	K	Т	F	K	S	A	V	K	Т	V	L	Н	T	A	K	K	A	Ι	S	S
34	Remove K1		W	K	S	F	L	K	Т	F	K	S	Α	K	K	Т	V	L	Н	Т	Α	L	K	Α	Ι	S	S
35	Remove K1, W2			K	S	F	L	K	T	F	K	S	A	K	K	T	V	L	Η	T	A	L	K	A	I	S	S
36	Remove S25, S26	K	W	K	S	F	L	K	T	F	K	S	A	K	K	T	V	L	Η	T	A	L	K	A	Ι		
37	Remove I24, S25, S26	K	W	K	S	F	L	K	T	F	K	S	A	K	K	T	V	L	Η	T	A	L	K	A			
38	non-polar face shuffle	K	I	K	S	AD		K	Т	L	K	S	F	K	K	Т	A	Α	Η	Т	L	F	K	V	W	S	5
39	polar face shuffle	S	W	S	K	F	L	K	K	F	Т	K	A	K	S	Η	V	L	Т	Т	A	L	S	Α	Ι	K	ŀ

In a preferred embodiment, the molecule (peptide) is helical in a hydrophobic environment. We have used circular dichroism spectroscopy to monitor α -helical structure in 50% trifluoroethanol, a mimic of the hydrophobic environment of the cytoplasmic membrane.

In an embodiment, successful peptides that are helical analogs with the desired biological activities have very little alpha-helical structure in benign medium (a non-denaturing medium like 50 mM PO₄ buffer containing 100 mM KCl, pH 7) monitored by circular dichroism spectroscopy. In an embodiment, this structural property can have importance in one or more of several potential mechanisms, for example: a) decreasing dimerization of molecule in benign medium (measured as described herein); b) allowing the peptide to more 55 easily penetrate through the cell wall to reach the membrane of the microbe. Furthermore, disruption of the α -helical structure in benign medium can still allow a positivelycharged peptide to be attracted to the negatively-charged cell wall surface of the microbe (e.g. lipopolysaccharide), but the 60 lack of structure can decrease the affinity of peptide for this surface which allows the peptide to more easily pass through the cell wall and enter the interface region of the membrane where the peptide is parallel to the surface of membrane. Here the peptide can be induced by the hydrophobic environment 65 of the membrane into its alpha-helical structure. In this alphahelical structure, we hypothesize that the non-polar face of

the peptide can interact with the hydrophobicity of the membrane, and its polar and positively-charged groups on the polar face can interact with the negatively-charged groups of the phospholipids on the surface of the membrane.

In an embodiment, a peptide is net positively-charged and amphipathic/amphiphilic when in an alpha-helical structure. For example, the alpha-helical peptide has a non-polar face or hydrophobic surface on one side of the molecule and a polar and positively-charged surface on the other side of the molecule; i.e., the molecule is amphipathic. Amphipathicity of the molecule can be calculated as described herein.

Certain peptide analogs were studied by temperature profiling in RP-HPLC from 5° C. to 80° C., to evaluate the self-associating ability of the molecules in solution. The ability to self-associate can be another important parameter in understanding peptide antimicrobial and hemolytic activities. It was generally found that a high ability to self-associate in solution was correlated with weak antimicrobial activity and strong hemolytic activity of the peptides. Biological studies showed that strong hemolytic activity of the peptides generally correlated with high hydrophobicity, high amphipathicity and high helicity. In most cases, the D-amino acid substituted peptides possessed an enhanced average antimicrobial activity compared with L-diastereomers. As illustrated herein, the therapeutic index of V_{681} was improved 90-fold and 23-fold against gram-negative and gram-positive bacteria, respectively (using geometric mean comparison). By replacing the central hydrophobic or hydrophilic amino acid

^{*}L-enantiomer unless otherwise indicated.

residue on the nonpolar or the polar face of these amphipathic molecules with a series of selected D- and L-amino acids, we further demonstrate that this method can be used for the rational design of other antimicrobial peptides with enhanced activities.

The antimicrobial peptides exemplified in the present invention contain amino acid substitutions in at least one of the two positions selected from the group represented as a and b in the peptide sequence given as KWKSFLKTFKaA bKTVLHTALKAISS (SEQ ID NO: 40) wherein a and b are selected from the group consisting of L-leucine, D-leucine, L-valine, D-valine, L-alanine, D-alanine, glycine, L-serine, D-serine, L-lysine, and D-lysine (L,A,S,V,K in L- and D-enantiomeric form and G). Particularly preferred antimicrobial peptides are referred to herein as NK_L, D-NK_D, NA_D, and D-NA_L and contain the following amino acid sequences. A most particularly preferred antimicrobial peptide is referred to herein as D-NK $_D$. In an alternative nomenclature, D-NK_D can be referred to as D-V13K_D.

Peptide name	Amino acid sequence (one letter code)
NK_L	$\begin{split} & \text{Ac-K}_{L} \text{-W}_{L} \text{-K}_{L} \text{-S}_{L} \text{-F}_{L} \text{-L}_{L} \text{-K}_{L} \text{-T}_{L} \text{-F}_{L} \text{-K}_{L} \text{-S}_{L} \text{-A}_{L} \frac{\left[\overline{K_{1}}\right]}{\left[K_{L}\right]} \text{-K}_{L} \text{-} \\ & \text{T}_{L} \text{-V}_{L} \text{-L}_{L} \text{-H}_{L} \text{-T}_{L} \text{-A}_{L} \text{-I}_{L} \text{-S}_{L} \text{-S}_{L} \text{-amide} \\ & (\text{SEQ ID NO:} 6) \end{split}$
${\tt D-NK}_D$	$\begin{split} & \text{Ac-K}_{\text{D}}\text{-W}_{\text{D}}\text{-K}_{\text{D}}\text{-S}_{\text{D}}\text{-F}_{\text{D}}\text{-L}_{\text{D}}\text{-K}_{\text{D}}\text{-T}_{\text{D}}\text{-F}_{\text{D}}\text{-K}_{\text{D}}\text{-S}_{\text{D}}\text{-A}_{\text{D}} \frac{\overline{ K_{\text{I}} }}{ K_{\text{D}} }\text{-K}_{\text{D}}\text{-K}_{\text{D}}\text{-L}_{\text{D}}\text{-K}_{\text{D}}\text{-L}_{\text{D}}\text{-K}_{\text{D}}\text{-L}_{\text{D}}\text{-S}_{\text{D}}\text{-S}_{\text{D}}\text{-smide} \\ & (\text{SEQ ID NO: 24}) \end{split}$
\mathtt{NA}_D	$\begin{array}{llllllllllllllllllllllllllllllllllll$
$\mathtt{D-NA}_L$	$ \begin{split} & \text{Ac-K}_D\text{-W}_D\text{-K}_D\text{-S}_D\text{-F}_D\text{-L}_D\text{-K}_D\text{-T}_D\text{-F}_D\text{-K}_D\text{-S}_D\text{-A}_D \boxed{\underline{A_D}} \text{-K}_D\text{-}\\ & \text{T}_D\text{-V}_D\text{-L}_D\text{-H}_D\text{-T}_D\text{-A}_D\text{-L}_D\text{-K}_D\text{-A}_D\text{-L}_D\text{-S}_D\text{-smide}\\ & (\text{SEQ ID NO:} 25) \end{split} $

Here a subscript D following an amino acid residue denotes 40 that the residue is a D-amino acid residue; similarly a subscript L denotes an L-amino acid residue. In the peptide name, an initial D- (not subscripted) denotes all D-amino acids in the peptide except where specified (e.g. D-NA, denotes all D-amino acids with the exception of a single substitution of 45 L-Ala in the center of the non-polar face specified by N). The boxed residues denote the differences at position 13 in the sequence which is in the center of the non-polar face (see FIG. 1).

in a larger peptide or protein. In an embodiment, a peptide of the invention is covalently or non-covalently associated with another composition. In a particular embodiment, said another composition is a polymer.

The peptides disclosed have antimicrobial activity against 55 a wide range of microorganisms including gram-positive and gram-negative bacteria. Detailed description of the microorganisms belonging to gram-positive and gram-negative bacteria can be found in Medical Microbiology (1991), 3^{rd} edition, edited by Samuel Baron, Churchill Livingstone, New 60 York. Examples of potentially susceptible bacteria include, but are not limited to, Escherichia coli, Salmonella typhimurium, Pseudomonas aeruginosa, Staphylococcus aureus, Staphylococcus epidermidis, Bacillus subtilis, Enterococcus faecalis, Corynebacterium xerosis, and Bacillus anthracis. 65 The antimicrobial activities of the inventive peptides have been demonstrated herein against the fore-mentioned gram-

positive and gram-negative bacteria because it is well known in the art that these bacteria are considered as model organisms for either gram-negative or gram-positive bacteria and thus any biological activity demonstrated again these model organisms are accepted as an indication of that demonstrated activity against the entire family of gram-negative or grampositive bacteria.

The antimicrobial peptides of the invention are useful as bactericides and/or bacteriostats for modification of infectivity, killing microorganisms, or inhibiting microbial growth or function and thus useful for the treatment of an infection or contamination caused by such microorganisms.

Also provided are therapeutic or otherwise active compositions suitable for human, veterinary, agricultural or pharmaceutical use, comprising one or more of the antimicrobial peptides of the invention and a suitable pharmaceutical carrier. Such therapeutic compositions can be formulated and administered as known in the art, e.g., for oral, parenteral or topical application for controlling and/or preventing infection 20 by a wide range of microorganisms including gram-positive and gram-negative bacteria.

In an embodiment of the invention, in vitro antimicrobial activity of these peptides demonstrated herein is an accurate predictor of in vivo antimicrobial activity.

Pharmaceutical compositions contain a therapeutically effective amount of one or more of the antimicrobial peptides and a suitable carrier. A therapeutically effective amount of an antimicrobial peptide can be readily determined according to methods well known in the art. For example, the amount will 30 vary depending on the severity of an infection, subject parameters such as the age and the size/weight of a subject with an actual or potential infection of a given microorganism, and the route of administration and the like.

The present invention relates to compositions comprising 35 one or more antimicrobial peptides of the invention in a microbicidal effective amount and a pharmaceutically acceptable carrier. Such compositions may additionally comprise a detergent. The addition of a detergent to such peptide compositions is useful to enhance antibacterial characteristics of the peptides. Although any suitable detergent may be used, the presently preferred detergent is a nonionic detergent such as Tween 20 or 1% NP40. Such antimicrobial pharmaceutical compositions can be formulated and administered in ways, as understood in the art for use local or systemic injection, for oral or topical application. In an embodiment, the antimicrobial peptides of the present invention can comprise from 0.0001% to 50% by weight of such compositions.

It will be understood that a composition for application, e.g. by systemic injection, will contain an antimicrobial pep-In an embodiment, a peptide of the invention is integrated 50 tide in a therapeutically effective amount or a therapeutically effective amount of an antimicrobial peptide can be conjugated to another molecule with specificity for the target cell type. The other molecule can be an antibody, ligand, receptor, or other recognition molecule. In an embodiment, the choice of the peptide is made with consideration of immunogenicity and toxicity for an actually or potentially infected host, effective dose of the peptide, and the sensitivity of the target microbe to the peptide, as known in the art.

> In an embodiment, the method of inhibiting the growth of bacteria using the peptides of the invention may further include the addition of one or more other antimicrobial agents (e.g. a conventional antibiotic) for combination or synergistic therapy. The appropriate amount of the peptide administered will typically depend on the susceptibility of a bacterium such as whether the bacterium is Gram negative or Gram positive, and will be easily discernable by one of ordinary skill in the

In an embodiment the invention also provides a composition that comprises the peptide, in an amount effective to kill a microorganism, and a suitable carrier. Such compositions may be used in numerous ways to combat microorganisms, for example in household or laboratory antimicrobial formulations using carriers well known in the art.

In an embodiment, the invention provides a peptide comprising SEQ ID NO:2. In an embodiment, the invention provides a peptide comprising a derivative of SEQ ID NO:1. In an embodiment, the invention provides a peptide selected from the group consisting of SEQ ID NO:2, 4-14, 16-25, and a derivative thereof. In an embodiment, a derivative comprises a substitution of at least one amino acid residue. In an embodiment, a derivative includes a substitution of at least 15 one amino acid residue; excepting SEQ ID NO:1. In an embodiment, a derivative comprises a truncation of at least one residue from an end. In an embodiment, a derivative comprises a truncation of at least two residues from an end. In an embodiment, a substitution replaces a hydrophilic residue 20 for a hydrophobic residue. In an embodiment, a substitution replaces a hydrophobic residue for a hydrophilic residue. In an embodiment, a substitution replaces a hydrophilic residue with a different hydrophilic residue. In an embodiment, a substitution replaces a hydrophilic residue with a different 25 hydrophilic residue. In an embodiment, a substitution replaces a residue with a different residue having a similar property, e.g., a polar side chain, a positively charged side chain, a negatively charged side chain, etc. In an embodiment, a substitution replaces an L-residue with a D-residue. In an 30 embodiment, a substitution replaces a D-residue with an L-residue. In an embodiment, all residues are D-residues.

In an embodiment, the invention provides a peptide comprising a sequence KWKSFLKTFKaAbKTVLHTALKAISS (SEQ ID NO:40) wherein a and b are selected from the group 35 consisting of L-leucine, D-Leucine, L-valine, D-valine, L-alanine, D-alanine, glycine, L-serine, D-serine, L-lysine, and D-lysine. In an embodiment, the peptide excludes SEQ ID NO:1.

In an embodiment, the invention provides peptide compositions as described herein, including fragments thereof; wherein the fragment length comprises a continuous stretch of at least about 14, at least about 17, at least about 20, at least about 23, at least about 24, or at least about 25 amino acids. In an embodiment, the invention provides a peptide composition wherein said composition is at least about 70%, at least about 80%, at least about 90%, or at least about 95% homologous to a sequence of a peptide described herein. In an embodiment, the invention provides a nucleic acid capable of encoding a peptide described herein. In an embodiment, a peptide of the 50 invention is intended to be exclusive of one or more of the following peptides: SEQ ID NO:1, SEQ ID NO:3, SEQ ID NO:15, and SEQ ID NO:26. In a particular embodiment, a peptide of the invention is exclusive of SEQ ID NO:1.

Where the peptides are to be used as antimicrobial agents, 55 they can be formulated, for example, in buffered aqueous media containing a variety of salts and buffers. Examples of the salts include, but are not limited to, halides, phosphates and sulfates, e.g., sodium chloride, potassium chloride or sodium sulfate. Various buffers may be used, such as citrate, 60 phosphate, HEPES, Tris or the like to the extent that such buffers are physiologically acceptable to the host that is being treated

Various excipients or other additives may be used, where the peptides are formulated as lyophilized powders, for subsequent use in solution. The excipients may include various polyols, inert powders or other extenders. 10

"Therapeutically effective" as used herein, refers to an amount of formulation, composition, or reagent, optionally in a pharmaceutically acceptable carrier, that is of sufficient quantity to ameliorate the state of the patient or animal so treated. "Ameliorate" refers to a lessening of the detrimental effect of the disease state or disorder in the recipient of the therapy. In an embodiment, a peptide of the invention is administered to a subject in need of treatment.

Pharmaceutically acceptable carrier preparations for administration include sterile or aqueous or nonaqueous solutions, suspensions, and emulsions. Examples of nonaqueous solvents are propylene glycol, polyethylene glycol, vegetable oils such as olive oil, and injectable organic esters such as ethyl oleate. Aqueous carriers include water, alcoholic/aqueous solutions, emulsions or suspensions, including saline and buffered media. Parenteral vehicles include sodium chloride solutions, Ringer's dextrose, dextrose and sodium chloride, lactated Ringer's, or fixed oils. Active therapeutic ingredients are often mixed with excipients that are pharmaceutically acceptable and compatible with the active ingredients. Suitable excipients include water, saline, dextrose, glycerol and ethanol, or combinations thereof. Intravenous vehicles include fluid and nutrient replenishers, electrolyte replenishers, such as those based on Ringer's dextrose, and the like. Preservatives and other additives may also be present such as, for example, antioxidants, chelating agents, and inert gases and the like. The actual dosage of the peptides, formulations or compositions containing such peptides can depend on many factors, including the size/weight, age, and health of an organism, however, one of ordinary skill in the art can use the following teachings and others known in the art describing the methods and techniques for determining clinical dosages (Spiker B., Guide to Clinical Studies and Developing Protocols, Raven Press, Ltd., New York, 1984, pp. 7-13, 54-60; Spiker B., Guide to Clinical Trials, Raven Press, Ltd., New York 1991, pp. 93-101; C. Craig. and R. Stitzel, eds., Modern Pharmacology, 2d ed., Little, Brown and Co., Boston, 1986, pp. 127-133; T. Speight, ed., Avery's Drug Treatment: Principles and Practice of Clinical Pharmacology and Therapeutics, 3d ed., Williams and Wilkins, Baltimore, 1987, pp. 50-56; R. Tallarida, R. Raffa and P. McGonigle, Principles in General Pharmacology, Springer-Verlag, new York, 1988, pp. 18-20) to determine the appropriate dosage to use.

In an embodiment, the following dosages are used: generally in the range of about 0.001 mg/kg to about 100 mg/kg and preferably from about 0.001 mg/kg to about 1 mg/kg final concentration are administered per day to an adult in any pharmaceutically acceptable carrier.

In another embodiment, the present invention may be used as a food preservative or in treating food products to control, reduce, or eliminate potential pathogens or contaminants. A peptide of the invention may be used as a disinfectant, for use in or with any product that must remain microbial free or be within certain tolerances. In an embodiment, treatment with a peptide provides at least partial regulation of infection or contamination.

In an embodiment it is also possible to incorporate or distribute the peptides within materials, on devices, or on objects (e.g. on an accessible surface), where microbial growth or viable presence is undesirable, as a method of microbicidal or microbistatic inhibition of microbial growth by administering to the devices or objects a microbicidal or microbistatic effective amount of peptide. In an embodiment, such devices or objects include, but are not limited to, linens, cloth, plastics, latex fabrics, natural rubbers, implantable devices, surfaces, or storage containers.

In an embodiment, the invention provides a method of disinfecting a surface of an article, said method comprising the step of applying to said surface an effective amount of a composition comprising at least one microbial peptide of the invention. In an embodiment, the invention provides a disinfecting solution comprising at least one microbial peptide of the invention and optionally an acceptable carrier.

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BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 illustrates the "host" peptide V_{681} (SEQ ID NO:1) as a helical wheel/nets and the sequences of the synthetic peptide analogs used in the present studies. Shown are the polar/hydrophilic face (boxed residues) and non-polar/hydrophobic face (circled residues). The hydrophobic face is indicated as an open arc, whilst the hydrophilic face is shown as a solid arc in the helical wheel. The substitution sites are at position 11 (triangled) on the polar face and position 13 (triangled) on the non-polar face. In the peptide sequences, X_L and X_D denote the L- or D-substituting amino acids. P denotes the polar face and N denotes the non-polar face. Ac denotes N_{α} -acetyl and amide denotes C_{α} -amide. One-letter codes are used for the amino acid residues.

FIG. 2 illustrates graphical results of Circular dichroism 25 (CD) spectra of peptides NL_D and NL_L (panel A) and peptides NK_D and NK_L (panel B) at pH 7 and 25° C., in 50 mM aq. PO4 containing 100 mM KCl. In both Panel A and Panel B, solid symbols represent the CD spectra of peptide analogs in benign buffer without TFE, whilst open symbols represent 30 CD spectra obtained in the presence of 50% TFE. The symbols used are: in Panel A, circle for NL_D and square for NL_L ; in Panel B, diamond for NK_D and triangle for NK_L .

FIG. 3 shows graphical results of Circular dichroism (CD) spectra of peptides V_{681} and D-V₆₈₁ (panel A), NK_L and 35 D-NK_D (panel B) and peptides NA_D and D-NA_L (panel C) at pH 7.4 and 5° C., in 50 mM aq. potassium phosphate buffer (KP buffer) containing 100 mM KCl. In Panels A, B and C, solid symbols represent the CD spectra of peptide analogs in KP buffer without trifluoroethanol (TFE), whilst open symbols represent CD spectra obtained in the presence of 50% TFE; the symbols used are: circles for L-peptides V_{681} , NK_L and NA_D and diamonds for D-peptides D-V₆₈₁, D-NK_L and D-NA_L.

FIG. 4 illustrates the CD temperature denaturation profiles $\,$ 45 of peptide $V_{681}.$ Panel A shows the change of V_{681} helical conformation over the temperature range from 5° C. to 80° C. in the hydrophobic medium. The experiment was carried out in 0.05% aq. TFA (pH 2) in the presence of 50% TFE. CD spectra at different temperatures are shown as different lines in the figure; Panel B: shows the stability plot of peptide V_{681} during CD temperature denaturation.

FIG. 5 illustrates the RP-HPLC temperature profiles of peptide V_{681} and its analogs. Conditions: RP-HPLC, narrowbore SB-C $_8$ column (150×2.1 mm ID; 5 µm particle size, 300 $\,^55$ Å pore size), linear A-B gradient (1% acetonitrile/min) at a flow-rate of 0.25 ml/min, where eluent A is 0.05% aqueous TFA and eluent B is 0.05% TFA in acetonitrile. Retention data has been collected in 3° C. increments within the temperature range from 5° C. to 80° C. Open symbols represent the temperature profiles of L-amino acid substituted peptides on either the non-polar or polar face of V_{681} (panels A and C); whereas solid symbols represent the temperature profiles of D-amino acid substituted peptides on either the non-polar or polar face of V_{681} (panels B and D). In all panels, the substituting amino acids used in either the non-polar or polar face of V_{681} are Val (circle), Leu (square), Ala (diamond), Ser (tri-

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angle), Lys (inverted triangle) and Gly (X). The temperature profile of the random coil control peptide (C1) is shown in the foure (+)

FIG. 6 illustrates the normalized RP-HPLC temperature profiles of peptide V_{681} and its analogs. Temperature profiles normalized to retention behavior of random coil peptide C1. Column and conditions: see FIG. 4. The retention behavior of the peptides was normalized to that of the random coil peptide C1 through the expression $(t_R^t - t_R^5)$ for peptides minus $(t_R^t - t_R^5)$ for C1), where $t_R^{\ t}$ are the retention times at a specific temperature of an antimicrobial peptide or the random coil peptide, and t_R^{5} are the retention times at 5° C. Open symbols represent the temperature profiling of L-amino acid substituted peptides on either the non-polar or polar face of V_{681} (panels A and C); whereas solid symbols represent D-amino acid substituted peptides on either the non-polar or polar face of V₆₈₁ (panels B and D). In all panels, amino acids used for substitution in either the non-polar or polar face of V_{681} are Val (circle), Leu (square), Ala (diamond), Ser (triangle), Lys (inverted triangle) and Gly (X).

FIG. 7 illustrates further peptide analogs including those generated by variations based on several techniques: the position of substitutions, nature of substitutions, multiple substitutions, truncations, and shuffling. FIG. 7A illustrates specific peptide sequences. FIG. 7B illustrates variants based on shuffling of the non-polar face residues. FIG. 7C illustrates variants based on shuffling of the polar face residues.

FIG. 8 illustrates the peptide retention behavior during RP-HPLC with increasing temperature. Column and conditions: RP-HPLC, SB-C₈ column (150×2.1 mm ID; 5 μm particle size, 300 Å pore size), linear A-B gradient (0.5% acetonitrile/min) at a flow-rate of 0.35 ml/min, where eluent A is 0.2% aqueous TFA and eluent B is 0.2% TFA in acetonitrile. The retention times of the peptides during the change of temperature are shown in Panel A. In Panel B, the retention time of peptides are normalized to 5° C. through the expression $(t_R^{t} - t_R^{-5})$, where t_R^{t} is the retention time at a specific temperature of an antimicrobial peptide or the random coil peptide, and t_R^5 is the retention time at 5° C. In Panel C, the retention behavior of the peptides was normalized to that of the random coil peptide C through the expression $(t_R^t - t_R^5)$ for peptides) minus $(t_R^t - t_R^5)$ for C). In Panels A, B and C, the symbols used are: circles for V_{681} and D- V_{681} , diamonds for NA_D and D- NA_L , squares for NK_L , D- NK_D and triangles for random coil peptide C.

FIG. 9 shows the results of peptide hemolysis of human red blood cells. Closed symbols were used for L-peptides and open symbols were used for D-peptides. Different symbols were used to represent the different concentrations of peptides in the hemolysis study.

FIG. 10 illustrates the peptide stability to proteolysis by trypsin. Closed symbols were used for L-peptides and open symbols were used for D-peptides. Circles denote V_{681} and D- V_{681} , squares denote NK_L and D- NK_D , and diamonds denote NA_D and D- NA_L .

FIG. 11 is a representation of the parent peptide NK_L and analogs (with different leucine substitutions) as helical nets showing the non-polar faces. The hydrophobic amino acid residues on the non-polar faces are boxed. The substituting leucine residues are colored red and the amino acid residues involved in the $i\rightarrow i+3$ and $i\rightarrow i+4$ hydrophobic interactions are colored yellow. The $i\rightarrow i+3$ and $i\rightarrow i+4$ hydrophobic interactions are shown as black bars and the numbers of hydrophobic interactions on the non-polar face are indicated. One-letter code is used for the amino acid residues.

FIG. 12 illustrates the relationship of peptide hydrophobicity and antimicrobial activity (MIC) and hemolytic activity

(MHC). Closed symbols and dotted line represent the relationship between hydrophobicity and hemolytic activity; open symbols and solid line represent the relationship between hydrophobicity and antimicrobial activity. Hydrophobicity was shown as the retention time of peptides in 5 RP-HPLC at room temperature. The lines were drawn to guide the eye.

FIG. 13 illustrates a space-filling model of peptides V681 and V13KL. Hydrophobic amino acids on the non-polar face of the helix are green; hydrophilic amino acids on the polar 10 face of the helix are grey; peptide backbone is colored white. The Lys substitution at position 13 (V13KL) on the non-polar face of the helix is blue. The models were created by the PyMOL v0.98.

DETAILED DESCRIPTION OF THE INVENTION

In general the terms and phrases used herein have their art-recognized meaning, which can be found by reference to standard texts, journal references and contexts known to those 20 skilled in the art. The following definitions are provided to clarify their specific use in the context of the invention.

When used herein, the term "amino acid" is intended to refer to any natural or unnatural amino acid, whether made naturally or synthetically, including any such in L- or D-con- 25 figuration. The term can also encompass amino acid analog compounds used in peptidomimetics or in peptoids. The term can include a modified or unusual amino acid or a synthetic derivative of an amino acid, e.g. diamino butyric acid and diamino propionic acid and the like.

The antimicrobial peptides of the invention are composed of amino acids linked together by peptide bonds. The peptides are in general in alpha helical conformation under hydrophobic conditions. Sequences are conventionally given from the amino terminus to the carboxyl terminus. Unless otherwise 35 noted, the amino acids are L-amino acids. When all the amino acids are of L-configuration, the peptide is said to be an L-enantiomer. When all the amino acids are of D-configuration, the peptide is said to be a D-enantiomer.

The term "minimal inhibitory concentration" (MIC) refers 40 to the lowest concentration of an antimicrobial agent (e.g., a peptide) required to prevent growth or otherwise modify a function of a microorganism under certain conditions, for example in liquid broth medium, and can be determined for a techniques well known in the art.

The term "minimal hemolytic concentration" (MHC) refers to the lowest concentration of an agent or peptide required to cause hemolysis of blood cells. MHC can be determined with red blood cells (RBC) from various species 50 including human red blood cells (hRBC).

The term "therapeutic index" (TI) is the ratio of minimal hemolytic concentration (MHC) over minimal inhibitory concentration (MIC) of an antimicrobial agent. Larger values generally indicate greater antimicrobial specificity.

The term "stability" can refer to an ability to resist degradation, to persist in a given environment, and/or to maintain a particular structure. For example, a peptide property of stability can indicate resistance to proteolytic degradation and to maintain an alpha-helical structural conformation.

The following abbreviations are used herein: A, Ala, Alanine; M, Met, Methionine; C, Cys, Cysteine; D, Asp, Aspartic Acid; E, Glu, Glutamic Acid; F, Phe, Phenylalanine; G, Gly, Glycine; H, H is, Histidine; I, Ile, Isoleucine; K, Lys, Lysine; L, Leu, Leucine; N, Asn, Asparagine; P, Pro, Proline; Q, Glu, 65 Glutamine; R, Arg, Arginine; S, Ser, Serine; T, Thr, Threonine; V, Val, Valine; W, Trp, Tryptophan; Y, Tyr, Tyrosine;

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RP-HPLC, reversed-phase high performance liquid chromatography; MIC, minimal inhibitory concentration; MHC, minimal hemolytic concentration; CD, circular dichroism spectroscopy; TFE, trifluoroethanol; TFA, trifluoroacetic acid; RBC, red blood cells; hRBC, human red blood cells.

The term "antimicrobial activity" refers to the ability of a peptide of the present invention to modify a function or metabolic process of a target microorganism, for example so as to at least partially affect replication, vegetative growth, toxin production, survival, viability in a quiescent state, or other attribute. In an embodiment, the term relates to inhibition of growth of a microorganism. In a particular embodiment, antimicrobial activity relates to the ability of an inventive peptide to kill at least one bacterial species. In a particular embodi-15 ment, the bacterial species is selected from the group consisting of gram-positive and gram-negative bacteria. In an embodiment, the term can be manifested as microbicidal or microbistatic inhibition of microbial growth.

The phrase "improved biological property" is meant to indicate that a test peptide exhibits less hemolytic activity and/or better antimicrobial activity, or better antimicrobial activity and/or less hemolytic activity, compared to the control peptide (e.g. V_{681}), when tested by the protocols described herein or by any other art-known standard protocols. In general, the improved biological property of the peptide is reflected in the therapeutic index (TI) value which is better that that of the control peptide.

The term "microorganism" herein refers broadly to bacteria, fungi, viruses, and protozoa. In particular, the term is applicable for a microorganism having a cellular or structural component of a lipid bilayer membrane. In specific embodiments, the membrane is a cytoplasmic membrane. Pathogenic bacteria, fungi, viruses, and protozoa as known in the art are generally encompassed. Bacteria can include gram-negative and gram-positive bacteria in addition to organisms classified in orders of the class Mollicutes and the like, such as species of the Mycoplasma and Acholeplasma genera. Specific examples of potentially sensitive gram-negative bacteria include, but are not limited to, Escherichia coli, Pseudomonas aeruginosa, Salmonella, Hemophilus influenza, Neisseria, Vibrio cholerae, Vibrio parahaemolyticus and Helicobacter pylori. Examples of potentially sensitive grampositive bacteria include, but are not limited to, Staphylococcus aureus, Staphylococcus epidermis, Streptonumber of different microorganisms according to standard 45 coccus agalactiae, Group A streptococcus, Streptococcus pyogenes, Enterococcus faecalis, Group B gram positive streptococcus, Corvnebacterium xerosis, and Listeria monocytogenes. Examples of potentially sensitive fungi include yeasts such as Candida albicans. Examples of potentially sensitive viruses include measles virus, herpes simplex virus (HSV-1 and -2), herpes family members (HIV, hepatitis C, vesicular, stomatitis virus (VSV), visna virus, and cytomegalovirus (CMV). Examples of potentially sensitive protozoa include Giardia.

"Therapeutically effective" as used herein, refers to an amount of formulation, composition, or reagent in a pharmaceutically acceptable carrier or a physiologically acceptable salt of an active compound, that is of sufficient quantity to ameliorate the undesirable state of the patient, animal, material, or object so treated. "Ameliorate" refers to a lessening of the detrimental effect of the disease state or disorder, or reduction in contamination, in the receiver of the treatment.

The peptides of the invention have antimicrobial activity by themselves or when covalently conjugated or otherwise coupled or associated with another molecule, e.g., polyethylene glycol or a carrier protein such as bovine serum albumin, so long as the peptides are positioned such that they can come

into contact with a cell or unit of the target microorganism. These peptides may be modified by methods known in the art provided that the antimicrobial activity is not destroyed or substantially compromised.

The invention may be further understood by the following 5 non-limiting examples.

EXAMPLE 1

Derivatives of Peptide V₆₈₁ with Modified Activity

In the studies described hereinafter, the 26-residue peptide having the sequence Ac-KWKSFLKTFKS-AVKTVLHTAL-KAISS-amide (V_{681} , SEQ ID NO:1) was utilized as the framework to study the effects of peptide hydrophobicity/ hydrophilicity, amphipathicity and helicity by one or more amino acid substitutions in the center of the polar and nonpolar faces of the amphipathic helix on biological activities. These studies demonstrate i) the importance of the peptide self-association parameter in the de novo design of amphip- 20 athic α-helical antimicrobial peptides; ii) that disruption of α-helical structure in benign conditions by D-amino acid substitutions or substitutions of hydrophilic/charged L-amino acids on the non-polar face can dramatically alter specificity; and iii) that these substitutions enhance antimicrobial activity, decrease toxicity and improve antimicrobial specificity while maintaining broad spectrum activity for gram-negative and gram-positive bacteria.

Peptide V_{681} , a 26-residue amphipathic antimicrobial peptide with a polar and non-polar face (28), was selected as the native parent peptide in this study (FIG. 1). Its polar face consists of 14 residues: six lysine residues, one histidine, four serines, and three threonines. In contrast, the non-polar face consists of 12 residues: three alanines, two valines, three leucines, two phenylalanines, one isoleucine and one tryp- 35 tophan residue. In this study, we chose D-/L-amino acid substitution sites at the center of the hydrophobic face (position 13) and at the center of the hydrophilic face (position 11) of the helix, such that these substitution sites were also located in the center of the overall peptide sequence. This was based 40 on our previous model peptide studies (26,31,34) that demonstrated that these central location substitutions had the greatest effect on peptide secondary structure. To study the effects of varying hydrophobicity/hydrophilicity on peptide biological activities, in the design of V_{681} analogs, five 45 L-amino acids (Leu, Val, Ala, Ser, Lys) and Gly were selected out of the 20 natural amino acids as the substituting residues. representing a wide range of hydrophobicity. The hydrophobicity of these six amino acid residues decreases in the order Leu>Val>Ala>Gly>Ser>Lys (26). Based on the relative 50 hydrophobicity of amino acid side-chains (26), leucine was used to replace the native valine on the non-polar face to increase peptide hydrophobicity and amphipathicity; alanine was selected to reduce peptide hydrophobicity/amphipathicity while maintaining high helicity; a hydrophilic amino acid, 55 serine, was selected to decrease the hydrophobicity/amphipathicity of V₆₈₁ in the non-polar face; positively-charged lysine was used to decrease further peptide hydrophobicity and amphipathicity. In contrast, the same amino acid substitutions on the polar face would have different effects on the 60 alteration of hydrophobicity/hydrophilicity and amphipathicity, since the native amino acid residue is serine on the polar face of V_{681} . As a result, on the polar face, leucine, valine and alanine were used to increase peptide hydrophobicity as well as decrease the amphipathicity of V_{681} , while lysine was selected to increase peptide hydrophilicity and amphipathicity. Previously, Kondejewski et al. (20, 35) and Lee et al. (25)

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successfully utilized D-amino acid substitutions to dissociate the antimicrobial activity and hemolytic activity of gramicidin S analogs. In the present study, D-enantiomers of the five L-amino acid residues were also incorporated at the same positions on the non-polar/polar face of $\rm V_{681}$ to change not only peptide hydrophobicity/hydrophilicity and amphipathicity but, more importantly, disrupt peptide helical structure. Since glycine does not exhibit optical activity and has no side-chain, the Gly-substituted analog was used as a reference for diastereomeric peptide pairs.

Since most peptide analogs were made based on a single amino acid substitution in either the polar or nonpolar faces of V_{681} , peptides were divided into two categories, N-peptides (nonpolar face substitutions) and P-peptides (polar face substitutions). Each peptide was named after the substituting amino acid residue, e.g., the peptide analog with L-leucine substitution on the nonpolar face of V_{681} is called NL_L . It is important to note that since the L-valine of the non-polar face and L-serine of the polar face are the original amino acid residues in the V_{681} sequence (FIG. 1), peptide analogs NV_L and PS_L are the same peptide as V_{681} .

A control peptide (peptide C) designed to exhibit negligible secondary structure, i.e., a random coil, was employed as a standard peptide for temperature profiling during RP-HPLC to monitor peptide dimerization. As shown in the previous study (29), this 18-residue peptide, with the sequence of Ac-ELEKGGLEGEKGGKELEK-amide (SEQ ID NO:26) clearly exhibited negligible secondary structure, even in the presence of the strong alpha-helix inducing properties of 50% trifluoroethanol (TFE) and at the low temperature of 5° C. ($[\theta]_{222}$ =-3,950).

To determine the secondary structure of peptides in different environments, circular dichroism (CD) spectra of the peptide analogs were measured under physiologically related pH and ionic strength (100 mM KCl, 50 mM aq. PO₄, pH 7 referred to as benign conditions) and also in 50% TFE to mimic the hydrophobic environment of the membrane. The native peptide, V₆₈₁, exhibited low alpha-helical content in benign conditions, i.e., $[\theta]_{222}$ of -12,900 compared to -27, 300 in 50% TFE, an increase in α -helical content from 45% to 94%, respectively (Table 3). From Table 3, in benign conditions, D-amino acid substituted peptides generally exhibited considerably less α-helical structure compared to their L-diastereomers. The negligible secondary structure characteristics of the D-peptides underlines the helix-disrupting properties of a single D-amino acid substitution, as demonstrated in our previous model α -helical peptide study (26). On the nonpolar face, the native L-Val residue was critical in maintaining α-helical structure. Substitution of L-Val with less hydrophobic amino acids (L-Ala, Gly, L-Ser and L-Lys) dramatically decreased the $\alpha\text{-helical}$ structure (NV $_L$, $[\theta]_{222}$ of –12,900 to values ranging from -1,300 to -3,450 for NS_L , NK_L , NG and NA_{τ}) (Table 3). Even the substitution with L-Ala, which is known to have the highest α -helical propensity of all 20 amino acids (34), could not stabilize the α -helical structure. This shows the importance of hydrophobicity on the nonpolar face in maintaining the α -helical structure. In contrast, substitution with a more hydrophobic amino acid (L-Leu for L-Val) on the non-polar face significantly increased α -helical structure ($[\theta]_{222}$ for peptide NL_L of -20,600 compared to peptide NV_Q of -12,900). It is noteworthy that, on the nonpolar face, the magnitude of the helical content of L-peptides in benign buffer was related to the hydrophobicity of the substituting amino acids, i.e., $NL_L > NV_L > NA_L > NS_L$, NK_L , again showing the importance of hydrophobicity on the nonpolar face in maintaining the α -helical structure. Due to their helix-disruptive ability, on the non-polar face, the D-amino

acid substitutions D-Val and D-Leu dramatically decreased α-helical structure in benign medium compared to their L-counterparts. However, whether L- or D-substitutions were made on the non-polar face, high helical structure could be induced by the hydrophobic environment of 50% TFE, a mimic of the membrane's hydrophobicity and α -helix inducing ability (Table 3). From Table 3, it is clear that, although D-amino acid substituted peptides were strongly induced into helical structure in 50% TFE, they were still generally less helical than the L-diastereomers, indicating that D-substitutions were still destabilizing of α-helical structure compared to their L-diastereomers in a hydrophobic environment.

non-polar face substitutions exhibited a greater range of molar ellipticity values in benign conditions than polar face analogs, demonstrating that the amino acid residues on the non-polar face of the helix played a more important role in peptide secondary structure than those on the polar face. As expected, Gly was destabilizing to α-helical structure whether on the non-polar or polar face due to its low α -helical propensity (34).

FIG. 2 shows the CD spectra of the most and the least hydrophobic substitutions on the non-polar face. In benign conditions, peptide NL_D showed much less helical structure than NL_L due to the helix-destabilizing ability of the D-amino

TABLE 3

	(Circular di	chroism d	lata of V68	1 peptide	analogs.		
		Ben	ign ^b			50%	TFE^d	
	X	e 1	Х	e D	X	e L	X	e D
Peptides ^a	$[\theta]_{222}$	% helix ^c						
NL	-20,600	71	-7,350	25	-28,250	98	-25,750	89
NV^f	-12,900	45	-2,800	10	-27,300	94	-26,000	90
NA	-3,450	12	-2,850	10	-28,950	100	-24,650	85
NS	-1,300	4	-1,700	6	-27,550	95	-22,200	77
NK	-1,450	5	-2,000	7	-26,250	91	-23,600	82
NG	-2,250	8	,		-24,350	84	,	
PL	-10,850	37	-2,950	10	-28,550	99	-26,100	90
PA	-13,600	47	-3,050	11	-27,600	95	-27,300	94
PS ^f	-12,900	45	-2,800	10	-27,300	94	-26,000	90
PV	-7,550	26	-2,400	8	-23,050	80	-20,800	72
PK	-5,950	21	-2,500	9	-27,350	94	-27,800	96
PG	-4,550	16	,		-25,950	90	,	

In this study, the L-substitutions on the polar face in benign 40 acid; whilst, in 50% TFE, both peptides could be induced to a medium had dramatically different effects on α-helical structure than the same substitutions on the non-polar face. For example, $NL_L([\theta]_{222} \text{ of } -20,600)$ differed from $PL_L([\theta]_{222} -$ 10,850), indicating that Leu stabilized α -helical structure on the non-polar face and destabilized α -helical structure on the polar face. Similarly, Val destabilized α -helical structure on the polar face; on the other hand, Ala and Ser destabilized helical structure on the non-polar face, whilst, Ala and Ser stabilized α -helical structure when substituted in the polar face, compared to the other amino acid substitutions. Taken together, even though Ala had the highest α-helical propensity of all amino acids (34), its α -helical propensity could not overcome the need for hydrophobicity on the non-polar face $([\theta]_{222}$ for peptides NA_L, -3,450 and NL_L, -20,600); 55 whereas, on the polar face, peptide PA_L exhibited high helical structure in benign ($[\theta]_{222}$ –13,600) in contrast to peptide PL_L $([\theta]_{222}$ -10,850) (Table 3). It is noteworthy that Val and Leu substitutions on the polar face decreased the amphipathicity of the helix as well as increased the hydrophobicity; however, 60 the lower helical content compared with the native PS, indicated that there should be a balance of amphipathicity and hydrophobicity to enhance the helical content. Similar to the substitutions on the non-polar face, all D-amino acid substitutions on the polar face were destabilizing to α -helical structure in benign medium; however, highly helical structure could be induced by adding 50% TFE. As shown in Table 3,

fully helical structure (FIG. 2, Panel A). In contrast, in benign condition, peptides NK_L and NK_D were random coils, due to the combined effects of decreasing hydrophobicity and amphipathicity by replacing the native L-Val to D-/L-Lys on the non-polar face; again, in 50% TFE, both of them were induced into highly helical structures, albeit that peptide NK, demonstrated slightly more helical content than peptide

Next, we prepared the enantiomeric peptides of V_{681} and analogs NK_L and NA_D . Peptides V_{681} and NK_L contain all L-amino acids and $\mathrm{D\text{-}V}_{681}$ and $\mathrm{D\text{-}NK}_D$ contain all D-amino acids. In the case of NA_D and D- NA_L , position 13 is D-alanine and L-alanine, respectively (Table 1). Thus, D-V₆₈₁, D-NK_D and D-NA_L are opposite in stereochemistry to the corresponding L-peptides, V_{681} , NK_L and NA_D , respectively. A control peptide C designed to exhibit negligible secondary structure, i.e., a random coil, was employed as a standard peptide for temperature profiling during RP-HPLC to monitor peptide dimerization (53, 19, 29).

To determine the secondary structure of the D-enantiomeric peptides in different environments, CD spectra of the peptide analogs were measured under benign conditions (100) mM KCl, 50 mM KH₂PO₄/K₂HPO₄, pH 7.4, referred to as KP buffer) and also in 50% trifluoroethanol (TFE) to mimic the hydrophobic environment of the membrane. As shown in FIG. 3, the parent peptide, V_{681} , was only partially helical in

Peptides are ordered by relative hydrophobicity to the native analog V_{681} at 5° C. N denotes non-polar face; P denotes polar face (FIG. 1)
The mean residue molar ellipticities, $[\theta]_{222}$, $(\deg, \operatorname{cm}^2, \dim \operatorname{ol}^{-1})$ at wavelength 222 nm were measured at 25° C. in benign buffer (100 mM KCl, 50 mM PO₄ pH 7.0).
The helical content (in percentage) of a peptide relative to the molar ellipticity value of the peptide NA_L in 50% trifluoroethanol (TFE).
The mean residue molar ellipticities ($\deg, \operatorname{cm}^2, \dim \operatorname{ol}^{-1}$) at wavelength 222 nm were measured at 25° C. in the benign buffer diluted 1:1 (v/v) with trifluoroethanol (TFE).
X_L and X_D denote the L- and D-substitutions, respectively.
N_V and PS_C are the same pentide, which is the native pentide V ...

 f_{NV_L} and PS_L are the same peptide, which is the native peptide V_{681}

KP buffer; peptides NK_L and NA_D exhibited negligible secondary structure in KP buffer due to disruption of the nonpolar face of the helix by introducing a hydrophilic L-lysine residue into peptide NK, or a helix-disruptive D-alanine residue into peptide NA_D . However, in the presence of 50% TFE, all three L-peptides were fully folded α -helical structures with similar ellipticities and helicity (Table 4). As expected, the D-peptides showed spectra that were exact mirror images compared to their L-enantiomers, with ellipticities equivalent but of opposite sign both in benign KP buffer and in 50% TFE $^{-10}$ (Table 4).

TABLE 4

		iro- icity ^b						
	t_{R5}	t _{R80}	Ве	nign	50%	6 TFE		
Peptide ^a	(min)	(min)	$[\theta]_{222}^c$	% helix ^d	$[\theta]_{222}^c$	% helix ^d	${\rm P_{\!\it A}}^e$	
V_{681}	96.6	88.8	-13000	46	-27950	99	7.2	
D-V ₆₈₁	96.6	88.8	13050	46	28100	100	7.2	
NA_D	81.2	73.2	-4000	14	-25000	89	4.1	
\widetilde{D} - \widetilde{NA}_L	81.2	73.2	4000	14	25050	89	4.1	
NK_L	74.9	64.7	-1400	5	-27000	96	2.1	
			1500	5	26950	96	2.1	

Peptide sequences are shown in Table 1.

Peptides are ordered by decreasing retention time (t_p) in RP-HPLC at pH 2 at temperatures of 5° C. and 80° C. which is a measure of overall hydrophobicity. The mean residue molar ellipticities, [6]₂₂₂, (deg. cm² .dmol²) at wavelength 222 nm were measured at 5° C. in benign conditions (100 mM KCl, 50 mM PO₄ pH 7.4) or in benign buffer containing 50% TFE by circular dichroism spectroscopy. The negative values in molar ellipticity denote the left-handed helices and the positive values denote the right-handed helices

ellipticity deflote the reference and the series of a peptide relative to the molar ellipticity value of the peptide D-V₆₈₁ in the presence of 50% trifluoroethanol (TFE). P₄ denotes the dimerization parameter of each peptide during the RP-HPLC temperature profiling, which is the maximal retention time difference of $(t_R^{-1}t_R^{-5}$ for peptide analogs) - $(t_R^{-1}t_R^{-5}$ for control peptide C)) within the temperature range, and $(t_R^{-1}t_R^{-5})$ is the retention time difference of a peptide at a specific temperature (t) compared with that at 5° C.

We next used temperature profiling during RP-HPLC to determine the self-association ability of the various analogs of V₆₈₁ which would occur through interaction of the nonpolar faces of these amphipathic α-helices. Using model 40 amphipathic α -helical peptides with all 20 amino acid substitutions in the center of the non-polar face, we showed previously that the model amphipathic peptides were maximally induced into an α-helical structure in 40% TFE and that the stability of the α -helix during temperature denaturation 45 was dependent on the substitution (26). In order to investigate the stability of V_{681} in a hydrophobic environment, we carried out a temperature denaturation study in solution, as monitored by circular dichroism spectroscopy. We used 50% aqueous TFE in 0.05% TFA to mimic the hydrophobic conditions 50 in the reversed-phase column since the hydrophobic environment of a reversed-phase column (hydrophobic stationary phase and the hydrophobic organic solvent in the mobile phase) could induce α -helical structure in a similar manner to TFE. FIG. 4, panel A shows the change of V_{681} helical conformation over the temperature range from 5° C. to 80° C. in the hydrophobic medium. At 5° C., 50% TFE induced full α -helical structure of V_{681} . During the temperature denaturation, the helical content of $V_{\rm 681}$ decreased with increasing temperature but even at 80° C. V_{681} remained significantly 60 $\alpha\text{-helical.}$ FIG. 4, panel B shows the stability profile of V_{681} with a transition temperature T_m of 79.3° C., where T_m is defined as the temperature when 50% of α -helical structure is denatured compared with the fully folded conformation of the peptide in 50% TFE at 5° C. These data support the view, that during temperature profiling in RP-HPLC, the peptides are fully helical at low temperatures such as 5° C. and can remain

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in the α-helical conformation at 80° C. in solution during partitioning in RP-HPLC. In addition, due to their hydrophobic preferred binding domains, the peptides will remain α -helical when bound to the hydrophobic matrix. Overall, these results indicate that V_{681} is a very stable α -helical peptide in a hydrophobic environment, whether it is in solution (such as 50% TFE), under the conditions of RP-HPLC or in the hydrophobic environment of the membrane.

It is well documented that the formation of a hydrophobic binding domain due to peptide secondary structure can affect peptide interactions with reversed-phase matrices, this effect having been observed especially for amphipathic α-helical peptides (26, 36-39). Indeed, Zhou et al. (39) clearly demonstrated that, because of this preferred binding domain, amphi-15 pathic α -helical peptides are considerably more retentive than non-amphipathic peptides of the same amino acid composition. In addition, the chromatography conditions characteristic of RP-HPLC (hydrophobic stationary phase, nonpolar eluting solvent) are able to induce and stabilize helical structure in potentially helical polypeptides (39-41) in a manner similar to that of the helix-inducing solvent TFE. From FIG. 1, it can be seen that the substitution site at position 13, in the center of the nonpolar face of the helix, ensures a maximal effect on the intimate interaction of the substituting sidechain with the reversed-phase stationary phase; thus, any differences in effective hydrophobicity via amino acid substitutions in the preferred binding domain can be readily monitored through consequent differences in RP-HPLC retention time.

The retention time data for the peptides is shown in Table 5 which records retention times at 5° C., the maximal retention times and retention times at 80° C. during the temperature profiling. Temperatures of 5° C. and 80° C. were the lower and upper temperature limits of temperature profiling in RP-HPLC, representing dimerization of the peptides at 5°C. and the monomerization of peptides at 80° C. due to dissociation of the dimers. The maximal retention times represent the threshold points at which peptides transform from dimeric to monomeric form. The retention profiles from 5-80° C. are shown in FIG. 5. Among the non-polar face substituted peptides, peptides with more hydrophobic substitutions (whether L- or D-amino acid substitutions) were more retained during RP-HPLC, i.e., peptides were eluted in the order of Lys, Gly, Ser, Ala, Val and Leu (Table 5). In addition, on the non-polar face, the L-analogs were always more retained than the D-diastereomers (Table 5, FIG. 5). Since the aforementioned preferred binding domain of amphipathic helices is actually the non-polar face of the helix, D-peptides had a smaller preferred binding domain compared with L-diastereomers, due to the helix disruptive ability of D-amino acids, resulting in lower retention times during RP-HPLC. In contrast, on the polar face, the elution order of peptides was not correlated with the order of amino acid side-chain hydrophobicity, e.g., PA_L and PS_L were more retained than PV_L (Table 5); PS_D was the most retained peptide among the D-amino acid substituted analogs on the polar face (Table 5). Indeed, on the polar face, peptides PL_L and PA_L , with the replacement of L-Ser by L-Leu or L-Ala, had increased overall hydrophobicity as revealed by higher retention times compared with V_{681} .

Although amino acid L-Val is much more hydrophobic than L-Ser, the observation that peptide PV_L was less retained than the native peptide V_{681} (with L-Ser at position 11 of the polar face) could be attributed to the helix-disrupting characteristics of the β -branched Val residue (also see Table 3). In contrast, at 80° C., PV_L became better retained than PS_L . Due to the unfolding of the helical structure at high temperature, the side-chain hydrophobicity of the substituting amino acid

TABLE 5 Relative hydrophobicity and association ability of peptide

analogs during RP-HPLC temperature profiling

in the peptide plays a more important role in the overall hydrophobicity. In a similar manner to the non-polar face substituted peptides, peptides with D-amino acids substituted into the polar face were dramatically less retained than their L-diastereomers. Due to the effect of the preferred binding 5 domain, peptides with substitutions on the non-polar face had a greater retention time range than those with polar face substitutions, e.g., 11.31 min for the L-peptides with nonpolar face substitutions versus 2.40 min for the L-peptides with polar face substitutions at 5° C., and 11.05 min versus 3.27 min for the D-peptides with non-polar or polar face substitutions, respectively, at 5° C. (Table 5).

The ability of the D-peptides to self-associate was determined by RP-HPLC temperature profiling. FIG. 8A shows the change in RP-HPLC retention times of the three pairs of 15 enantiomers and the control peptide C over a temperature range of 5° C. to 80° C. As expected, L- and D-peptide enantiomers were totally inseparable over this temperature range, since each pair of peptides is identical in sequence and must adopt identical conformations on interacting with the 20 reversed-phase matrix, whether in an all-L- or all-D-conformation. RP-HPLC retention behavior has been frequently utilized to represent overall peptide hydrophobicity (53,26). In the present study, the hydrophobicity of the three peptide pairs is in the order $V_{681}/D-V_{681}>NA_D/D-NA_L>NK_L/D-25$ NK_D (Table 4), which agrees with the change in hydrophobicity of the substitutions at position 13 in order of the most hydrophobic to the least hydrophobic amino acid residue (Val in V_{681} >Ala in NA>Lys in NK) (54). FIG. **8**B shows the retention behavior of the peptides after normalization of the 30 retention times to the corresponding retention time at 5° C. in order to highlight differences in the elution behavior of peptides as the temperature is increased from 5° C. to 80° C. For example, the retention times of peptides $V_{681}/D-V_{681}$ increase with increasing temperature (up to ~30° C.) followed 35 by a retention time decrease with a further temperature increase. Such a temperature profile is characteristic of a peptide exhibiting self-association (53, 29, 19). As shown in FIG. 8C, the peptide self-association parameter, P₄, represents the maximum change in peptide retention time relative 40 to the random coil peptide C. Since peptide C is a monomeric random coil peptide in both aqueous and hydrophobic media, its retention behavior over the temperature range 5° C. to 80° C. represents only general temperature effects on peptide retention behavior, i.e., a linear decrease in peptide retention 45 time with increasing temperature due to greater solute diffusivity and enhanced mass transfer between the stationary and mobile phases at higher temperatures (55). Thus, after normalization to the retention times of peptide C, the retention behavior of the peptides represents only peptide self-associa- 50 tion ability. Note that the higher the P_A value, the greater the self-association ability. The order of peptide self-association ability of the three pairs of peptide enantiomers is identical to the order of peptide hydrophobicity, i.e., V₆₈₁/D-V₆₈₁ have the highest dimerization ability in solution among the three 55 pairs of peptide enantiomers (P_4 =7.2, Table 4); in contrast, NA_D/D - NA_L showed a weaker ability to self-associate when compared to $V_{681}/D-V_{681}$ ($P_A=4.1$, Table 4); $NK_L/D-NK_D$ exhibited the lowest dimerization ability (P_A =2.1, Table 4). From Table 4, it is also clear that the peptide retention times 60 at 80° C. are dramatically lower than those at 5° C. Apart from the decrease in retention time due to the general temperature effects noted above, unraveling of the α -helix will also occur with increasing temperature, resulting in the loss of the nonpolar face of the amphipathic α -helical peptides and, hence, 65 reduced retention times as the peptides become increasingly random coils.

		$t_R (min)^l$	·			
Peptides ^a	5° C.	Max	80° C.	5° C.	80° C.	$\Pr_A \pmod{d}$
Non-polar ^e	_		_	$\Delta t_R (X-N)$	V_L) $(\min)^c$	
NL_L	48.16	50.45	47.02	0.06	1.19	4.22
NV_L^f	48.10	49.99	45.83	0	0	3.63
NA_L	43.94	44.88	41.38	-4.16	-4.45	2.59
NS_L	40.72	41.08	37.62	-7.38	-8.21	1.82
NK_L	36.85	36.91	33.22	-11.25	-12.61	1.10
NG	39.96	40.22	36.74	-8.14	-9.09	1.64
NL_D	45.10	46.37	43.03	-3.00	-2.80	3.02
NV_D	42.55	43.43	40.15	-5.55	-5.68	2.63
NA_D	40.49	41.01	38.00	-7.61	-7.83	2.19
NS_D	37.02	37.12	34.08	-11.08	-11.75	1.46
NK_D	34.05	34.05	30.96	-14.05	-14.87	1.10
Polar ^e	_		_	$\Delta t_R (X-PS)$	S_L) $(\min)^c$	
PL_L	48.78	51.23	47.51	0.68	1.68	4.33
PA_L	48.25	50.57	46.63	0.15	0.80	4.15
PS_L^f	48.10	49.99	45.83	0	0	3.63
$\widetilde{PV_L}$	47.83	49.93	46.18	-0.27	0.35	3.91
PK_L	46.38	47.90	43.89	-1.72	-1.94	3.17
PG	45.86	47.09	43.07	-2.24	-2.76	2.82
PS_D	45.47	46.60	42.59	-2.63	-3.24	2.73
PA_D	45.19	46.36	42.57	-2.91	-3.26	2.82
PL_D	44.73	45.85	42.14	-3.37	-3.69	2.82
$\overline{PV_D}$	42.96	43.83	40.51	-5.14	-5.32	2.54
PK_D	42.20	42.87	39.29	-5.90	-6.54	2.23
C _g	22.74	_	18.64	_	_	_

 $^{\rm o}$ Peptides are ordered by relative hydrophobicity to the native L-Val substituted analog on the non-polar face and L-Ser substituted analog on the polar face. $^{\rm o}$ denotes the retention times at 5° C., the maximal retention times and the retention times at

be denotes the retention times at 5° C., the maximal retention times and the retention times at 80° C. during the temperature profiling. Genotes the difference of retention time relative to that of the native peptide V_{681} (NV_L for the non-polar face substitutions and PS_L for the polar face substitutions), representing the relative hydrophobicity of the peptide analogs. $^{4}P_{A}$ denotes the association parameter of each peptide during the RP-HPLC temperature profiling, which is the maximal retention time difference of ($(t_R^2 + t_R^2)$ for peptide analogs)—($(t_R^2 + t_R^2)$ for ontrol peptide C)) within the temperature range, and $(t_R^2 + t_R^2)$ is the retention time difference of a peptide at a specific temperature (t) compared with that at 5° C. 'denotes the amino acid substituted on either the non-polar face (N) or the polar face (P) of the annihibation rative peptide V_{681} (see FIG. 1).

the amphipathic native peptide V_{681} (see FIG. 1). NV_L and PS_L are the same peptide, which is the native peptide V_{681} .

 g Peptide C is a random coil control used to calculate P_{A} values, see footnote d.

Elution times during RP-HPLC have frequently been utilized as a measure of relative hydrophobicity of peptide analogs (26,31). In the current study, peptide analogs differed only by a single amino acid substitution on either the nonpolar face or the polar face of V_{681} ; thus, the retention time data in Table 5 can be considered to reflect the hydrophobicity difference between peptide analogs. In order to more easily visualize the variation in hydrophobicity of the peptide analogs, the retention time data in Table 5 were normalized relative to that of the native peptide V_{681} at $5^{\rm o}$ C. and $80^{\rm o}$ C., respectively. Hydrophobicity relative to the native peptide V_{681} indicates an increase or decrease of the apparent peptide hydrophobicity with the different amino acid substitutions on the polar or non-polar face. Again, from Table 5 and FIG. 5, for non-polar face substituted peptides, there was a wide range of peptide hydrophobicity in the order L-Leu>L-Val>L-Ala>L-Ser>Gly>L-Lys at both 5° C. and 80° C. On both the non-polar and polar faces, the relative hydrophobicities of the D-peptides was always less than their L-diastereomers, indicating the helix-disrupting characteristic of D-amino acids also leads to disruption of the preferred binding domain of the helices. On both non-polar and polar faces, peptides exhibited a greater retention time range at 80° C. than at 5° C., also indicating that, due to the unfolding of the helical structures at 80° C., the side-chain hydrophobicity of

the substituted amino acids played a more essential role in determining the overall hydrophobicity of the peptide analogs.

The hydrophobicity/hydrophilicity effects of substitutions on the non-polar face relative to the native peptide V_{681} were large. For example, NV $_Q$ to NA $_L$, to NS $_A$, and to NK $_L$ resulted in decreases in hydrophobicity of -4.45, -8.21 and -12.61 min at 80° C., respectively (Table 5). In fact, the same substitutions, i.e., PV $_L$ to PA $_L$, to PS $_L$, and to PK $_L$, resulted in overall hydrophobicity changes of the peptide by +0.45, -0.35 and -2.29 min at 80° C., respectively. This indicates that the polar face substitutions affected overall hydrophobicity of the peptide in a minor way relative to substitutions on the non-polar face. In fact, the effect was of 10 times less for Ala, >20 times less for Ser and >5 times less for Lys.

Since its introduction, the technique of RP-HPLC temperature profiling has been applied on several types of molecules, including cyclic β -sheet peptides (30), monomeric α -helices and α -helices that dimerize (29), as well as α -helices that dimerize to form coiled-coils (42). Although peptides are 20 eluted from a reversed-phase column mainly by an adsorption/desorption mechanism (43), even a peptide strongly bound to a hydrophobic stationary phase will partition between the matrix and the mobile phase when the acetonitrile content becomes high enough during gradient elution. 25 The proposed mechanism of action for temperature profiling of α -helical peptides in RP-HPLC has been explained in detail by Mant, et al. (29). In summary, the mechanism is based on four assumptions: (i) at low temperature, just as an amphipathic α -helical peptide is able to dimerize in aqueous solution (through its hydrophobic, nonpolar face), it will dimerize in solution during partitioning in reversed-phase chromatography; (ii) at higher temperatures, the monomerdimer equilibrium favors the monomer as the dimer is disrupted; (iii) at sufficiently high temperatures, only monomer 35 is present in solution; and (iv) peptide is always bound in its monomeric helical form to the hydrophobic stationary phase, i.e., the dimer can only be present in solution and disruption of the dimer is required for rebinding to the RP-HPLC matrix.

It is well accepted that the amphipathicity of antimicrobial 40 peptides is necessary for their mechanism of action, since the positively-charged polar face will help the molecules reach the biomembrane through electrostatic interaction with the negatively-charged head groups of phospholipids, and then the nonpolar face of the peptides will allow insertion into the 45 membrane through hydrophobic interactions, causing increased permeability and loss of barrier function of target cells (6,7). Thus, we believe that peptide self-association (i.e., the ability to dimerize) in aqueous solution is a very important parameter to understand antimicrobial activity. If the self-association ability of a peptide in aqueous media is too strong (forming dimers and burying the non-polar face), it could decrease the ability of the peptide to dissociate and penetrate into the biomembrane and to kill target cells.

FIG. 5 shows the temperature profiling of all L-/D-amino 55 acid substituted peptide analogs during RP-HPLC from 5° C. and 80° C. As mentioned above, the dimerization is temperature-dependent. At low temperatures, peptides exist in a dimer-monomer equilibrium during RP-HPLC partitioning, with the dimeric unbound state favored and dissociation 60 required for rebinding; thus, the retention times are relatively low. With the increase of temperature, equilibrium is shifted toward the monomeric form in solution due to the disruption of the dimer. The higher solution concentration of monomer during partitioning increases the on-rate for the bound state, 65 and the retention time therefore increases. It should be noted that the increased temperature also introduces other general

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effects on retention time because of lower mobile phase viscosity and a significant increase in mass transfer between the stationary phase and mobile phase. These effects decrease retention time with increasing temperature in a linear fashion, as shown for the random coil control peptide C (FIG. 5). Conversely, for the dimerized peptides, at a given temperature dimers are disrupted and converted to monomers and the retention time reaches the maximal value. Above this critical temperature, one will observe a decrease in retention time with increasing temperature because of the low mobile phase viscosity and increase in mass transfer. In addition, the above described temperature-induced conformational changes, as monitored by CD, may also have an impact by decreasing the retention time with increasing temperature, largely due to the destabilization of peptide α -helical structure and loss of preferred binding domain at high temperatures. To eliminate these general effects during RP-HPLC, the data from FIG. 5 were normalized relative to the temperature profile of the random coil peptide standard C, and normalized to the retention time at 5° C., the latter of which is presented as a dotted line in FIG. 6.

It was observed that the peptide analogs in this study showed dramatic varying dimerization ability in solution (FIG. 6). The maximal values of the change of retention times $((t_R^t - t_R^5 \text{ for peptide}) - (t_R^t - t_R^5 \text{ for C})) \text{ in FIG. 6}$ were defined as the peptide association parameter (P_A) to quantify the association ability of peptide analogs in solution (Table 5). As seen from the data in Table 5, peptides with higher relative hydrophobicity generally showed stronger self-association ability in solution. The P_A values of the peptide with nonpolar face substitutions were of the same order as their relative hydrophobicity, indicating that the hydrophobicity on the hydrophobic face of the amphipathic helix was essential during dimerization, since the dimers are formed by the binding together of the non-polar faces of two amphipathic molecules. In contrast, the different relationship between P_{\perp} and the relative hydrophobicity of the peptides with polar face substitutions demonstrated that the hydrophobicity on the polar face of the helices plays a less important role in peptide association. Generally speaking, the P₄ values of L-peptides were significantly greater than those of their D-diastereomers, indicating the importance of helical structure during dimerization (Table 5). In Table 5, in most cases, the peptides with polar face substitutions had greater P_A values than the corresponding peptide analogs with the same amino acid substitutions on the non-polar face. This is exactly what one would expect since polar face substitutions have little effect on the preferred dimerization domain, whereas non-polar face substitutions would dramatically affect the hydrophobicity and dimerization ability of the peptide.

Amphipathicity of the L-amino acid substituted peptides was determined by the calculation of hydrophobic moment (32) using the software package Jemboss version 1.2.1 (33), modified to include the hydrophobicity scale determined in our laboratory (see the Examples section for details) (Table 6). Peptide amphipathicity, for the non-polar face substitutions, was directly correlated with side-chain hydrophobicity of the substituted amino acid residue, i.e., the more hydrophobic the residue the higher the amphipathicity (values of 6.70 and 5.60 for NL_L and NK_L , respectively); in contrast, on the polar face, peptide amphipathicity was inversely correlated with side-chain hydrophobicity of the substituted amino acid residue, i.e., the more hydrophobic the residue, the lower the amphipathicity (compare PK_L and PL_L with amphipathicity values of 6.62 and 5.45, respectively, Table 6).

The native sequence, V_{681} was very amphipathic with a value of 6.35. To place this value in perspective, the sequence

of V_{681} can be shuffled to obtain an amphipathic value of 0.96 (KHAVIKWSIKSSVKFKISTAFKATTI, SEQ ID NO: 41) or a maximum value of 8.10 for the sequence of HWSKLLKS-FTKALKKFAKAITSVVST (SEQ ID NO:42). The range of amphipathicity values achieved by single substitutions on the polar and non-polar faces varied from a low of 5.45 for PL_{I} to a high of 6.70 for NL_{τ} (Table 6). Even though single substitutions changed the amphipathicity, all the analogs remained very amphipathic, e.g., even with a lysine substitution on the non-polar face, NK_L has a value of 5.60.

TABLE 6

	Amphipathicity of	of peptide ana	logs.
Peptide	Amphipathicity ^a	Peptide	Amphipathicity ^a
NL_L	6.70	PL_L	5.45
NV_L^{b}	6.35	PV_L	5.82
NA_L	5.98	PA_L	6.21
NG	5.85	PG	6.35
NS_L	5.85	$PS_L^{\ b}$	6.35
NK,	5.60	PK_{I}	6.62

Amphipathicity was determined by the calculation of hydrophobic moment (31) using hydrophobicity coefficients determined by reversed-phase chromatography (see Materials and Methods for details). Peptides NV $_L$ and PS $_L$ are the same peptide as V $_{68L}$

Concerning the mechanism of action of antimicrobial peptides, many models have been proposed, among which the "barrel-stave" mechanism and the "carpet" model are two (44). In brief, the "barrel-stave" mechanism describes the formation of transmembrane channels/pores by bundles of amphipathic α -helices, as their hydrophobic surfaces interact with the lipid core of the membrane and the hydrophilic surfaces point inward, producing an aqueous pore (45); in contrast, the "carpet" model was proposed for the first time to 35 describe the mechanism of action of dermaseptin S (46), describing the contact of antimicrobial peptides with the phospholipid head group throughout the entire process of membrane permeation which occurs only if there is a high local concentration of membrane-bound peptide. The major 40 difference between the two mechanisms is, in the carpet model, peptides lie at the interface with their hydrophobic surface interacting with the hydrophobic component of the lipids but are not in the hydrophobic core of the membrane, and neither do they assemble the aqueous pore with their 45 hydrophilic faces. A NMR study has shown that the cyclic β-sheet peptide analog of gramicidin S lavs in the interface region parallel with the membrane where its hydrophobic surface interacts with the hydrophobic fatty acyl chains and the positively charged residues can still interact with the 50 negatively charged head groups of the phospholipids (47).

Whichever the mechanism, the prerequisite step is the attraction of the peptide molecule to the membrane, followed by insertion into the bilayer. If peptide molecules self-associate in aqueous solution, peptides with lower self-associat- 55 ing ability in an aqueous medium can more easily penetrate into the lipid membrane. From Table 5 and Table 6, peptides with higher relative hydrophobicity on their non-polar face created higher amphipathicity and generally showed stronger self-associating ability in solution; in contrast, for peptides 60 with polar face substitutions, increasing hydrophobicity lowers amphipathicity yet the peptides still strongly self-associate, which indicates that peptide amphipathicity plays a less important role in peptide self-association when changes in amphipathicity are created on the polar face. In addition, 65 self-associating ability is correlated with the secondary structure of peptides, i.e., in this study, disrupting the peptide

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helical structure by replacing the L-amino acid with its D-amino acid counterpart decreases the P₄ values (Table 4

The hemolytic activity of the peptides against human erythrocytes was determined as a major measure of peptide toxicity toward higher eukaryotic cells. As mentioned before, the native peptide V_{681} (also named as NV_Q or PS_L) had strong hemolytic activity, with a minimal hemolytic concentration (MHC value) of 15.6 µg/ml (Table 7). In this study, due to the alteration of hydrophobicity, amphipathicity and stability, the hemolytic activity of the best variants of peptide V_{681} was significantly decreased to no detectable activity, a >32 fold decrease for NK_L (Table 7).

From Table 7, it is clear that, for the non-polar face substi-15 tuted peptides, hemolytic activity was correlated with the side-chain hydrophobicity of the substituting amino acid residue, i.e., the more hydrophobic the substituting amino acid, the more hemolytic the peptide, consistent with our previous study on the β -sheet antimicrobial peptide gramicidin S (22). 20 For example, the MHC of peptide NL_T was 7.8 μ g/ml; in contrast, the MHC was decreased, parallel with the reduction of hydrophobicity, to an undetectable level for peptide NK_L. Peptide hydrophobicity and amphipathicity on the non-polar face were also correlated with peptide self-associating ability, thus peptides with less self-association in benign conditions also exhibited less hemolytic activity against eukaryotic cells. In contrast, for polar face substituted peptides, the relationships between self-association, hydrophobicity/amphipathicity and hemolytic activity were less clear. Of course, the hydrophobic non-polar face remained very similar when L-substitutions were made on the polar face; thus, dimerization and hydrophobicity of the non-polar face would be less affected and hemolytic activity would remain relatively

In addition to hydrophobicity/amphipathicity, peptide helicity seemed to have an additional effect on hemolytic activity. In general, on both the non-polar and polar faces, D-amino acid substituted peptides were less hemolytic than their L-diastereomers. For example, NA, had a MHC value of 31.2 μ g/ml compared to NA_D with a value of 250 μ g/ml, an 8-fold decrease in hemolytic activity. Similarly, PV_L had a MHC value of 7.8 μ g/ml compared to PV_D with a value of 125 μg/ml, a 16-fold decrease in hemolytic activity (Table 7). This phenomenon generally correlated with peptide self-associating ability, since D-diastereomeric analogs exhibited weaker self-associating ability than L-analogs. Additionally, D-substitutions disrupt helicity which, in turn, disrupts hydrophobicity of the non-polar face. This result was also consistent with the data of Shai and coworkers (23,24), who demonstrated that, through multiple D-amino acid substitutions, the helicity of peptides is substantially reduced leading to decreased hemolytic activity. Thus, peptide structure is important in the cytotoxicity towards mammalian cells although these disturbed helices can still maintain antibacterial activity.

As shown in Table 7, peptide analogs with non-polar face substitutions exhibited a greater range of hemolytic activity $(7.8 \,\mu\text{g/ml})$ to not detectable) than the polar face substitutions (4 to 125 μg/ml), again indicating that the non-polar face of the helix may play a more essential role during the interaction with the biomembrane of normal cells. As expected, the peptides with the polar face substitutions showed stronger hemolytic activity than the peptides with the same amino acid substitutions on the non-polar face, which may be attributed to the different magnitude of the hydrophobicity change by the same amino acid substitutions on different sides of the amphipathic helix. Interestingly, in this study, all polar face

substituted peptides except $\mathrm{PL}_{\mathcal{D}}, \mathrm{PV}_{\mathcal{D}}$ and $\mathrm{PK}_{\mathcal{D}}$ showed stronger hemolysis of erythrocytes than $\mathrm{V}_{681};$ in contrast, on the non-polar face, only peptides $\mathrm{NL}_{\mathcal{D}}$ and $\mathrm{NL}_{\mathcal{L}}$ were more hemolytic than $\mathrm{V}_{681}.$

The antimicrobial activity of the peptides with either nonpolar face or polar face amino acid substitutions against a range of gram-negative microorganisms is shown in Table 7. The geometric mean MIC values from 6 microbial strains in this table were calculated to provide an overall evaluation of antimicrobial activity against gram-negative bacteria. It is apparent that many peptide analogs showed considerable improvement in antimicrobial activity against gram-negative bacteria over the native peptide V_{681} , e.g., peptides NK_L and PK_D exhibited 2.8-fold and 3.4-fold improvement on the average MIC value compared to V_{681} , respectively (geometric mean comparison; Table 7). Generally, the peptide analogs have high activity against bacterial strains of *E. coli* (UB 1005 wt and DC2 abs), *S. typhimurium* C610 abs and *P. aeruginosa* H187 wt in this study (Table 7).

For gram-negative bacteria, disruption of peptide helicity out weighted other factors in the improvement of antimicro- 20 bial activity; i.e., in most cases, the peptides with D-amino acid substitutions showed better antimicrobial activity than L-diastereomers. The exceptions were peptides NS_D and NK_D . The reason for the low activity of peptides NS_D and NK_D was possibly the combined effects of the destabilization of the helix, the decrease of hydrophobicity on the non-polar face and the disruption of amphipathicity, highlighting the importance of maintaining a certain magnitude of hydrophobicity and amphipathicity on the non-polar face of the helix for biological activity, i.e., perhaps there is a combined threshold of helicity and hydrophobicity/amphipathicity required for biological activity of α-helical antimicrobial peptides. In this study, peptide self-associating ability (relative hydrophobicity) seemed to have no general relationship to MIC; however, interestingly, for peptides with L-hydrophobic amino acid substitutions (Leu, Val and Ala) in the 35 polar and non-polar faces, the less hydrophobic the substituting amino acid, the more active the peptide against gram-

Table 8 shows the antimicrobial activity of the peptides against gram-positive microorganisms. By introducing D-/L- 40 amino acid substitutions, we improved the antimicrobial activity of peptide V_{681} against gram-positive bacteria by as much as 2.7-fold (geometric mean MIC values for V_{681} were $6.3 \,\mu\text{g/ml}$ compared to $2.3 \,\mu\text{g/ml}$ for PS_D, Table 8). Compared with peptide V_{681} , most of the peptide analogs with increased antimicrobial activity against gram-positive microorganisms were D-amino acid substituted peptides (6 D-peptides versus 1 L-peptide). It was surprising to observe that peptides with polar face substitutions showed an overall greater improvement in MIC than those with non-polar face substitutions. Generally speaking, increasing the hydrophobicity of the 50 native peptide V_{681} by amino acid substitutions at either the polar or the non-polar face decreased the antimicrobial activity against Gram-positive bacteria, e.g., peptides NL_L , PL_L , PV_L and PA_L (Table 8). Amino acid substitutions of D-Ser and D-Lys on the non-polar face significantly weakened the activity, in a similar manner to the anti-Gram-negative activity, indicating again the importance of maintaining a certain magnitude of helicity, hydrophobicity/amphipathicity on the nonpolar face of the helix for peptide Gram-positive antimicrobial activity.

Therapeutic index is a widely employed parameter to represent the specificity of antimicrobial reagents. It is calculated by the ratio of MHC (hemolytic activity) and MIC (antimicrobial activity); thus, larger values in therapeutic index indicate greater antimicrobial specificity. As mentioned above, the native peptide V_{681} is a peptide with good antimicrobial activity coupled with strong hemolytic activity; hence, its therapeutic index is low (1.8 and 2.5 for gram-negative and

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gram-positive bacteria, respectively) and comparable to general toxins like melittin. In this study, by altering peptide hydrophobicity/hydrophilicity, amphipathicity and helicity, we significantly increased the therapeutic index of peptide $V_{\rm 681}$ against gram-negative bacteria by 90-fold (Table 7) and Gram-positive bacteria by 23-fold (Table 8). As indicated in Tables 7 and 8, there was a greater range of therapeutic indices for peptides with the non-polar face substitutions compared with the polar face substitutions, which was consistent with peptide self-association studies, indicating that the non-polar face of the helix may play a more important role in the mechanism of action.

Table 9 summarizes the data for these peptide analogs with improved therapeutic index values relative to the native peptide V₆₈₁. From Table 3 and Table 9, it is clear that all peptides with improved therapeutic indices are those showing less stable helical structure in benign medium (either the D-amino acid substituted peptides or the hydrophilic amino acid substituted peptides on the non-polar face). The peptide with the best therapeutic index among all the analogs is NK_L with a 90-fold improvement compared with V_{681} against Gramnegative bacteria; in contrast, peptide NA_D is the analog with the broadest specificity against all the tested gram-negative and gram-positive microorganisms, having a 42-fold improvement in therapeutic index against gram-negative bacteria and a 23-fold improvement against gram-positive bacteria. It is noteworthy that the hemolytic activity of these two peptides was extremely weak; in addition, peptides NK, and NA_D exhibited improved antimicrobial activity compared to peptide V_{681} against gram-negative bacteria and identical antimicrobial activity against gram-positive bacteria.

In order to evaluate the biological activities of the D-enantiomeric peptides, namely D-NA_L and D-NK_D, the hemolytic activity of the peptides against human erythrocytes was determined as the maximal peptide concentration that produces no hemolysis after 18 hours of incubation at 37° C. The parent peptide, V₆₈₁, exhibited the strongest hemolytic activity with a value of 7.8 μ g/ml, compared to peptides NK_L and NA_D (250.0 μ g/ml and 31.3 μ g/ml, respectively) (Table 10). The activity of the D-enantiomers was quantitatively equivalent to that of L-enantiomers.

In the studies described above (see Tables 5-7), the hemolytic activities of peptides V681, NA_D and NK_L were determined as 15.6 µg/ml, 250.0 µg/ml and greater than 250.0 μg/ml, respectively, following incubation for 12 hours at 37° C. instead of the 18 hours incubation used for the D-enantiomers. Hence, a hemolysis time study was carried out to investigate the extent of hemolysis during different periods of incubation and different peptide concentrations in order to explore the relationship between hemolysis and incubation time. As shown in FIG. 9, a hemolysis time study was carried out over 8 hours at peptide concentrations of 8, 16, 32, 64, 125, 250 and 500 μg/ml. The percent hemolysis of cells was determined spectrophotometrically by comparison with the complete hemolysis of cells in water. From FIG. 9, peptides V_{681} , $D-V_{681}$, NA_D and $D-NA_L$ exhibited an increase in erythrocyte hemolysis with increasing incubation time at most peptide concentrations. Significantly, peptides NK_L and D-NK_D showed negligible (less than 10%) or no hemolysis against human red blood cells after 8 hours even at the extremely high peptide concentration of 500 µg/ml. From FIG. 9, it is clear that the enantiomeric peptide pair V_{681}/D - ${
m V}_{
m 681}$ exhibit cell lysis to a significantly greater extent than the NA_D/D-NA_L peptide pair. Thus, 100% hemolysis was already apparent after just 1 hour incubation for V_{681} and $D\text{-}V_{681}$ at a peptide concentration of 500 µg/ml, whilst this value was approximately 40% for peptides NA_D and $D-NA_L$. In addition, while a small increase in hemolysis was apparent for V_{681} and D- V_{681} at a peptide concentration of just 8 µg/ml, negligible hemolysis was seen for NA_D and D-NA_L at a peptide concentration of 32 µg/ml. It is interesting to note that for

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our lead compounds, peptides $\mathrm{NA}_D/\mathrm{D}\text{-}\mathrm{NA}_L$ and NK_L/D - NK_D , D-enantiomers generally exhibited slightly weaker hemolytic activity compared to L-enantiomers at most peptide concentrations, the only exception being NA_D and D- NA_L at 500 µg/ml, where the activity of the L-peptide is greater at less than 4 hours incubation but weaker from 4 hours onwards.

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Pseudomonas aeruginosa strains used in this study are a diverse group of clinical isolates from different places in the world. Antibiotic susceptibility tests show that these Pseudomonas aeruginosa strains share similar susceptibility to most antibiotics except that there is about a 64-fold difference for the range of ciprofloxacin susceptibility (Vasil, M. L. et al unpublished data). Antimicrobial activities of peptide enantiomers against Pseudomonas aeruginosa strains are shown in Table 10. In general, the antimicrobial activity of Land D-enantiomers against Pseudomonas aeruginosa varied within 4-fold. CP204, a Pseudomonas aeruginosa strain from cystic fibrosis patients, is apparently the most susceptible strain to these peptides with MIC values from 7.8 to 31.3 μg/ml. The geometric mean MIC values from six Pseudomo- 20 nas aeruginosa strains were calculated to provide an overall evaluation of antimicrobial activity of these peptides against Pseudomonas aeruginosa. Fold improvement was calculated by comparing the geometric mean MIC value of each peptide to that of the parent peptide V_{681} . It is clear that, in most cases, all-D-peptides exhibited slightly better antimicrobial activity than their L-enantiomers.

The therapeutic indices of the peptides against Pseudomonas aeruginosa are shown in Table 10. By replacing L-valine with D-alanine or L-lysine, we significantly increased the therapeutic index against Pseudomonas aeruginosa strains 30 by 4.7 fold (peptide NA_D) and 16.7 fold (peptide NK_L). In addition, by making enantiomeric D-peptides, we further improved the therapeutic index against Pseudomonas aeruginosa by 7.3 fold (peptide D-NA_L) and 26.7 fold (peptide D-NK_D) as measured by the standard microtiter dilution 35 method. Peptide D-NK_D showed no hemolysis at 500 μg/ml after 8 hours in our time study (FIG. 9) which is a more stringent test of hemolytic activity. The hemolytic activities of V₆₈₁ were not different when determined by the standard microtiter dilution method at 18 hours or by the time study at 8 hours using a peptide concentration of 500 µg/ml (Table 10 and FIG. 9). Thus, using a peptide concentration of 500 μ g/ml, the MHC value of D-NK_D increased the fold improvement of the therapeutic index compared to V_{681} to 53-fold. Our overall conclusion is that the key differences in biological activities lie in the much more dramatic variations in

hemolytic activity (MHC values) since the differences in antimicrobial activity between the six peptides are relatively minor (Table 10). Thus, the dramatic increases in therapeutic indices arise from the difference in hemolytic activity.

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Table 11 shows peptide antimicrobial activity and therapeutic indices against six different Gram-negative bacterium strains. It is clear to see that, in general, all-L-peptides have equivalent antimicrobial activity against Gram-negative bacteria compared to their D-enantiomers, since MIC values of L- and D-enantiomers are all within a 2-fold difference. In Table 11, geometric mean MIC was calculated to provide an overall view of antimicrobial activity of the peptides against Gram-negative bacteria and was also used to calculate the therapeutic index. Peptides NK_L and D-NK_D both show a 40-fold improvement in therapeutic index against Gramnegative bacteria compared to peptide V_{681} as measured by the standard microtiter dilution method. For this improvement in the therapeutic index of peptides against Gram-negative bacteria, peptide hemolytic activity is again the key factor. Using our most stringent test of hemolytic activity, the MHC value of 500 μ g/ml for peptide D-NK_D showed no hemolysis after 8 hours in our time study (FIG. 9) and the fold improvement of the therapeutic index compared to V681 was 80-fold.

The antimicrobial activity against Gram-positive bacteria and a fungus of L- and D-enantiomeric peptides is compared in Table 12. It is clear that our peptides were effective in killing the Gram-positive bacteria tested. The least sensitive Gram-positive bacterium to the enantiomeric peptides was E. faecalis. The effectiveness of the peptides to lyse cells of the fungus C. albicans was similar to E. faecalis. For Grampositive bacteria and the fungus, D-peptides again generally showed the same or better antimicrobial activity than L-enantiomers. Although the geometric mean MIC values of peptides NK_L/D - NK_D were among the two lowest, due to the high MHC values (poor or no hemolytic activity), the therapeutic indices of NK_L and D- NK_D were the highest against Grampositive bacteria and the fungus. Similar to that seen against Gram-negative bacteria, the therapeutic index of peptide D-NK_D was increased 34.7-fold and 16.7-fold against Grampositive bacteria and C. albicans, respectively, due to the poor or no hemolytic activity as measured by the standard microtiter dilution method. Using our most stringent test of hemolytic activity, the MHC value of 500 µg/ml for peptide D-NK_D showed no hemolysis after 8 hours in our time study (FIG. $\overline{9}$) and the fold improvement of the therapeutic indices compared to V₆₈₁ were 69- and 33-fold against Gram-negative bacteria and fungus, respectively.

TABLE 7

Antimicrobial (MIC) and hemolytic (MHC) activities of peptide analogs against gram-negative bacteria and human red blood cells.

				MIC ^a (μg/ml)					
Peptides	E. coli UB1005 wt ^e	E. coli DC2 abs ^e	S. typhimurium C587 wt ^e	S. typhimurium C610 abs ^e	P. aeruginosa H187 wt ^e	P. aeruginosa H188 abs ^e	GM^b	MHC ^c (µg/ml) hRBC	Therapeutic index ^d
NL_L	6.4	5.0	32.0	10.1	12.7	32.0	12.7	7.8	0.6
NV_L^f	7.1	4.5	20.2	5.7	6.4	20.2	8.8	15.6	1.8
NA_L	2.5	2.5	6.4	2.5	5.0	6.4	3.8	31.2	8.1
NG	2.5	2.5	5.0	2.5	6.4	10.1	4.1	125.0	30.2
NS_L	2.5	2.5	6.4	2.0	6.4	10.1	4.2	125.0	30.1
NK_L^g	2.5	1.6	4.0	1.3	8.0	5.0	3.1	>250.0	163.0
NL_D	3.2	2.5	16.0	3.2	6.4	10.1	5.5	7.8	1.4
NV_D	2.5	1.6	5.0	2.0	4.0	8.0	3.3	62.5	19.0
NA_D^g	1.6	2.0	5.0	2.0	4.0	10.1	3.3	250.0	75.7
NS_D	3.2	2.0	12.7	2.0	18.3	20.2	6.3	>250.0	79.9
NK_D	3.2	2.5	32.0	1.0	32.0	25.4	7.7	>250.0	65.0

TABLE 7-continued

Antimicrobial (MIC) and hemolytic (MHC) activities of peptide analogs against gram-negative bacteria and human red blood cells.											
PL_L	16.0	5.0	32.0	12.7	20.2	32.0	16.6	4.0	0.2		
$\overline{\mathrm{PV}_L}$	6.4	4.0	32.0	5.0	10.1	20.2	9.7	7.8	0.8		
PA_L^-	6.4	4.0	20.2	4.0	10.1	16.0	8.3	15.6	1.9		
PG	5.0	2.5	12.7	3.2	4.0	10.1	5.2	7.8	1.5		
PS_L^f	7.1	4.5	20.2	5.7	6.4	20.2	8.8	15.6	1.8		
PK_L	10.1	4.0	25.4	8.0	25.4	32.0	13.7	4.0	0.3		
PL_D^-	5.0	2.5	10.1	3.2	4.0	10.1	5.0	31.2	6.2		
$\overline{\text{PV}_D}$	5.0	2.5	10.1	4.0	6.4	16.0	6.1	125.0	20.5		
PA_D	4.0	2.5	8.0	2.0	5.0	8.0	4.3	15.6	3.6		
PS_D^-	2.5	1.6	5.0	1.6	2.0	10.1	2.9	15.6	5.3		
PK_D	3.2	1.6	3.2	1.6	2.0	6.4	2.6	31.2	11.8		

^aAntimicrobial activity (minimal inhibitory concentration) is given as the geometric mean of three sets of determinations.

TABLE 8

Antimicrobial (MIC) and hemolytic (MHC) activities of peptide analogs against gram-positive bacteria and human red blood cells.

MI	C^a
(μg/	ml
R	011

Peptides	S. aureus 25923 wt ^e	S. aureus SAP0017 methR ^e	S. epidermidis C621 wt ^e	B. subtilis C971 wt ^e	E. faecalis C625 wt ^e	C. xerosis C875 wt ^e	GM^b	MHC ^c (µg/ml) hRBC	Therapeutic index ^d
$\overline{\mathrm{NL}_L}$	32.0	25.4	8.0	3.2	32.0	2.5	10.9	7.8	0.7
NV_L^f	16.0	9.0	5.0	2.2	16.0	2.5	6.3	15.6	2.5
NA_L	8.0	5.0	3.2	2.0	16.0	2.0	4.5	31.2	6.9
NG	25.4	10.1	3.2	2.0	50.8	2.0	7.4	125.0	16.8
NS_L	16.0	12.7	4.0	2.5	50.8	1.6	7.4	125.0	16.9
NK_L^g	64.0	64.0	5.0	1.6	64.0	1.3	11.8	>250.0	42.3
NL_D	5.0	4.0	2.5	2.5	6.4	1.6	3.3	7.8	2.4
$\overline{\mathrm{NV}_D}$	4.0	3.2	1.6	1.3	12.7	1.3	2.8	62.5	22.7
NA_D^g	8.0	5.0	2.0	1.6	32.0	1.6	4.3	250.0	57.8
NS_D	64.0	64.0	12.7	2.5	64.0	2.0	16.0	>250.0	31.3
NK_D	64.0	64.0	25.4	3.2	64.0	2.0	18.7	>250.0	26.8
PL_L	32.0	32.0	16.0	5.0	50.8	2.5	14.8	4.0	0.3
PV_L	16.0	12.7	8.0	2.5	20.2	1.3	6.9	7.8	1.1
PA_L	16.0	12.7	4.0	2.5	20.2	2.0	6.6	15.6	2.4
PG	8.0	5.0	4.0	2.0	12.7	2.0	4.5	7.8	1.7
PS_L^f	16.0	9.0	5.0	2.2	16.0	2.5	6.3	15.6	2.5
PK_L	32.0	16.0	6.4	3.2	32.0	4.0	10.5	4.0	0.4
PL_D	8.0	5.0	4.0	2.0	16.0	2.0	4.7	31.2	6.7
$\overline{\text{PV}_D}$	16.0	8.0	4.0	2.5	32.0	2.0	6.6	125.0	19.0
PA_D	6.4	5.0	2.5	2.0	12.7	1.6	3.9	15.6	4.1
PS_D^-	4.0	2.5	2.0	1.3	6.4	1.0	2.3	15.6	6.7
PK_D	4.0	2.5	2.0	2.0	12.7	1.0	2.8	31.2	11.0

^aAntimicrobial activity (minimal inhibitory concentration) is given as the geometric mean of three sets of determinations.

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^bGM denotes the geometric mean of MIC values from all 6 microbial strains in this table.

[&]quot;Hemolytic activity (minimal hemolytic concentration) was determined on human red blood cells (hRBC). When no detectable hemolytic activity was observed at

^{250.0} µg/ml, a value of 500 µg/ml was used for calculation of the therapeutic index.

^aTherapeutic index = MHC (µg/ml)/geometric mean of MIC (µg/ml). Larger values indicate greater antimicrobial specificity.

^ewt denotes the wild type strain and abs denotes the antibiotics sensitive strain.

 $f_{\rm NV_L}$ and PS_L are the same peptide, which is the native peptide V_{681} .

gThe boxed results show the two best peptides with broad spectrum activity in terms of the therapeutic index against both Gram-negative and Gram-positive bacteria.

 $[^]b\mathrm{GM}$ denotes the geometric mean of MIC values from all 6 microbial strains in this table.

 $^{^{\}circ}$ Hemolytic activity (minimal hemolytic concentration) was determined on human red blood cells (hRBC). When no detectable hemolytic activity was observed at 250.0 μ g/ml, a value of 500 μ g/ml was used for calculation of the therapeutic index.

 $[\]label{eq:def-def} {}^d\text{The rapeutic index} = \text{MHC } (\mu g/\text{ml})/\text{geometric mean of MIC } (\mu g/\text{ml}). \text{ Larger values indicate greater antimicrobial specificity.}$

 $^{^{}e}$ wt denotes the wild type strain and methR denotes the methicillin-resistant strain.

 $^{{}^}f\!\mathrm{NV}_L$ and PS_L are the same peptide, which is the native peptide V_{681} .

gThe boxed results show the two best peptides with broad spectrum activity in terms of the therapeutic index against both Gram-negative and Gram-positive bacteria.

TABLE 9

Effect of amino acid substitutions on the biological activity of ${ m V_{681}}^a$.											
		Gram	ı-negative			Gram-positive					
Peptide	MIC ^b (μg/ml)	Fold ^e	Therapeutic index ^c	Fold ^e	$\begin{array}{c} {\rm MHC}^d \\ ({\rm \mu g/ml}) \end{array}$	Fold ^e	MIC ^b (μg/ml)	Fold ^e	Therapeutic index ^c	Fold	
V ₆₈₁	8.8	1.0	1.8	1.0	15.6	1.0	6.3	1.0	2.5	1.0	
NA_L	3.8	2.3	8.1	4.5	31.2	2.0	4.5	1.4	6.9	2.8	
NG	4.1	2.1	30.2	16.8	125.0	8.0	7.4	0.9	16.8	6.7	
NS_L	4.2	2.1	30.1	16.7	125.0_	8.0	7.4	0.9	16.9	6.8	
NK_L^f	3.1	2.8	163.0	90.6	>250.0	32.1	11.8	0.5	42.3	16.9	
$\overline{\mathrm{NV}_D}$	3.3	2.7	19.0	10.6	62.5	4.0	2.8	2.3	22.7	9.1	
NA_D^f	3.3	2.7	75.7	42.1	250.0	16.0	4.3	1.5	57.8	23.1	
NS_D	6.3	1.4	79.9	44.4	>250.0	32.1	16.0	0.4	31.3	12.5	
NK_D	7.7	1.1	65.0	36.1	>250.0	32.1	18.7	0.3	26.8	10.7	
PL_D	5.0	1.8	6.2	3.4	31.2	2.0	4.7	1.3	6.7	2.7	
PV_D	6.1	1.4	20.5	11.4	125.0	8.0	6.6	1.0	19.0	7.6	
PA_D	4.3	2.0	3.6	2.0	15.6	1.0	3.8	1.7	4.1	1.6	
PS_D	2.9	3.2	5.3	2.9	15.6	1.0	2.3	2.7	6.7	2.7	
PK_D	2.6	3.4	11.8	6.6	31.2	2.0	2.8	2.3	11.0	4.4	

 $[^]a$ Only the peptide analogs with a therapeutic index greater than $m V_{681}$ are included in this table.

TABLE 10

	Hemolytic activities of peptides and MIC against Pseudomonas aeruginosa Hemolytic Antimicrobial activity activity MHC MIC (µg/ml) ^c											trains. Therapeutic index		
Peptide	$(\mu g/ml)^a$	Fold^b	PAO 1	WR 5	PAK	PA 14	M 2	CP 204	GM^d	Fold^e	MHC/MIC ^f	Fold ^g		
V ₆₈₁	7.8	1	15.6	62.5	31.3	62.5	15.6	15.6	27.8	1.0	0.3	1.0		
D-V ₆₈₁	7.8	1	15.6	31.3	31.3	15.6	31.3	31.3	24.8	1.1	0.3	1.0		
NA_D	31.3	4	15.6	62.5	62.5	31.3	7.8	7.8	22.1	1.3	1.4	4.7		
$\overline{\text{D-NA}_L}$	31.3	4	7.8	31.3	31.3	7.8	15.6	7.8	13.9	2.0	2.2	7.3		
NK,	250.0	32	31.3	250.0	125.0	62.5	31.3	7.8	49.6	0.6	5.0	16.7		
D-NK _D Ciprofloxacin ⁱ	250.0 ^h	32	15.6 4	62.5 1	125.0 4	15.6 8	31.3 8	15.6 64	31.2	0.9	8.0	$26.7 (53.4)^h$		

MHC denotes the maximal peptide concentration that produces no hemolysis after 18 hours in the standard microtiter dilution method.

 $V_{681} \\ D-V_{681}$

 NA_D D- NA_L

7.8

7.8

31.3

31.3

6.3

6.3

3.1

3.1

3.1

3.1

1.6

3.1

12.5

12.5

12.5

12.5

3.1

6.3

3.1

3.1

TABLE 11 Antimicrobial (MIC) and hemolytic (MHC) activities of peptide analogs against Gram-negative bacteria

				anc	i humar	red blo	od cells	S.				
	$\frac{\mathrm{MHC}^b}{(\mu\mathrm{g/ml})}$											
	MHC ^α (μg/ml)	E. coli C498	E. coli C500		St C610	<i>Pa</i> H187	<i>Ра</i> Н188			Therapeut	ic index	
Peptides	hRBC	wt ^c	abs^c	wt^c	abs^c	wt^c	abs ^c	GM^d	Folde	MHC/MIC	Fold ^g	

6.3

6.3

3.1

3.1

3.1

6.3

1.6

3.1

5.0

6.3

3.1

3.9

1.0

0.8

1.6

1.3

1.6

1.2

10.1

8.0

1.0

0.8

6.3

5.0

^bAntimicrobial activity (minimal inhibitory concentration) was given as the geometric mean data of Tables IV & V.

Therapeutic index = MHC (in μg/ml)/MIC (in μg/ml). Larger values indicate greater antimicrobial specificity.

^dHemolytic activity (minimal hemolytic concentration) was determined on human red blood cells (hRBC). When no detectable hemolytic activity was observed at 250.0 μ g/ml, a value of 500 μ g/ml was used for calculation of the therapeutic index and fold increased. edenotes the fold improvement in activity compared with the corresponding data of the native peptide V_{681} .

^fThe boxed results show the two best peptides with broad spectrum activity in terms of the therapeutic index against both Gram-negative and Gram-positive bacteria.

 $[^]b\mathrm{Denotes}$ the fold decrease in hemolytic activity compared to the parent peptide $\mathrm{V}_{681.}$

^eMIC is minimal inhibitory concentration that inhibited growth of six *Pseudomonas aeruginosa* strains. MIC is given based on three sets or more of determinations. "Denotes the geometric mean of a diverse group of $Pseudomonas\ aeruginosa\ clinical\ isolates.$

 $[^]e$ Denotes the fold improvement in antimicrobial activity (geometric mean data) compared to the parent peptide V_{681} .

fTherapeutic index is the ratio of the MHC value (µg/ml) over the geometric mean MIC value (µg/ml). Large values indicate greater antimicrobial specificity. g Denotes the fold improvement in therapeutic index compared to the parent peptide V_{681} .

 $^{^{}h}$ In the hemolysis time study, D-V13K $_{D}$ showed no hemolysis at 500 μ g/ml at 8 hours, which would increase the fold improvement to 53.4 fold in the therapeutic

index. Relative susceptibility of these Pseudomonas strains to ciprofloxacin with the number 1 denoting the most susceptible strain.

TABLE 11-continued

Antimicrobial (MIC) and hemolytic (MHC) activities of peptide analogs against Gram-negative bacteria and human red blood cells.

		MHC^b (µg/ml)									
	MHC ^a (μg/ml)	E. coli C498	E. coli C500	<i>St</i> C587	St C610	<i>Ра</i> Н187	<i>Ра</i> Н188		· -	Therapeu	tic index ^f
Peptides	hRBC	wt^c	abs^c	wt^c	abs^c	wt^c	abs ^c	GM^d	Fold ^e	MHC/MIC	Fold ^g
NK_L D-NK _D	250.0 250.0 ^h	3.1 3.1	3.1 3.1	6.3 6.3	3.1 3.1	6.3 6.3	3.1 3.1	3.9 3.9	1.3 1.3	64.1 64.1	40.1 40.1 (80.2) ^h

 $^{^{}o}$ Hemolytic activity (maximal peptide concentration with no hemolysis after 18 hours) was determined on human red blood cells (hRBC).

TABLE 12

	Antimicrol	oial (MIC	and hemo	olytic (MHC) a	ctivities o MIC ^b (µg/ml		nalogs agair	st Gran	n-positiv	e bacteria	, fungus a	ind human red l	blood cells	S
	MHC^a	S. aureus	S. aureus	S. epidermidis	B. subtilis	E. faecalis	C. xerosis			Thera ind	L	MIC^b		peutic lex ^f
Peptides	(μg/ml) hRBC	622 wt ^c	623 methR ^c	C621 wt ^c	C971 wt ^c	C625 wt ^c	C875 wt ^c	GM^d	Fold^d	MHC/ MIC	Fold ^g	(μg/ml) C. albicans	MHC/ MIC	Fold ^g
V_{681}	7.8	3.1	6.3	6.3	3.1	6.3	3.1	4.4	1.0	1.8	1.0	12.5	0.6	1.0
$D-V_{681}$	7.8	6.3	6.3	1.6	1.6	6.3	3.1	3.5	1.3	2.2	1.2	6.3	1.2	2.0
NA_D	31.3	1.6	3.1	1.6	3.1	12.5	1.6	2.8	1.6	11.2	6.2	12.5	2.5	4.2
$D-NA_L$	31.3	3.1	3.1	1.6	1.6	12.5	1.6	2.8	1.6	11.2	6.2	12.5	2.5	4.2
NK_L	250.0	12.5	12.5	6.3	3.1	50.0	1.6	7.9	0.6	31.6	17.6	50.0	5.0	8.3
D-NK _E	250.0^{h}	6.3	6.3	3.1	1.6	12.5	1.6	4.0	1.1	62.5	34.7	25.0	10.0	16.7

^aHemolytic activity (maximal peptide concentration with no hemolysis after 18 hours) was determined on human red blood cells (hRBC)

FIG. 10 illustrates the stability of L- and D-enantiomeric peptides in the presence of trypsin at a molar ratio of 20,000:1 50 or 20 mM:1 µM (peptide:trypsin) at 37° C. The assay was carried out for 8 hours; however, only the stability of the peptides in the first 60 minutes is shown in FIG. 10, due to the reason that all three L-peptides were 100% digested by trypsin after 1 hour of incubation at 37° C. It is interesting to 55 see that peptide V_{681} showed a slightly more resistance against trypsin than NK_L and NA_D ; in contrast, NK_L and NA_D generally showed a similar trend of degradation in the presence of trypsin. All three peptides were 86-97% digested after 30 minutes. In sharp contrast, D-peptides were completely 60 stable (no degradation) to trypsin even at a 10-fold higher molar concentration ratio of trypsin versus peptide (2 mM peptide: 1 µM trypsin) for 8 hours.

As mentioned previously, the "barrel-stave" and the "carpet" mechanisms are the two main theories used to explain the 65 mechanism of action of antimicrobial peptides. However, neither mechanism alone can fully explain the data in this

study. For example, the hemolytic activity is correlated to the peptide hydrophobicity and amphipathicity on the non-polar face, which may be consistent with the "barrel-stave" mechanism, i.e., peptides interact with the hydrophobic core of the membrane by their non-polar face to form pores/channels. In contrast, the antimicrobial activity is not correlated with peptide hydrophobicity/amphipathicity, showing that the "barrel-stave" mechanism may not be suitable to explain the mechanism of antimicrobial action. Indeed, the "carpet" mechanism may best explain the interaction between the peptides and the bacterial membrane. Based on the above observations, we propose that both mechanisms are in operation for the peptides used in this study, i.e., the mechanism depends upon the difference in membrane composition between prokaryotic and eukaryotic cells. If the peptides form pores/channels in the hydrophobic core of the eukaryotic bilayer, they would cause the hemolysis of human red

 $(69.4)^{1}$

 $(33.4)^h$

^bAntimicrobial activity (minimal inhibitory concentration) is from three sets of determinations.

 $[^]cE$. coli denotes $Escherichia\ coli, St$ denotes $Salmonella\ typhimurium, Pa\ denotes\ Pseudomonas\ aeruginosa,\ wt\ denotes\ the\ wild-type\ lab$

strain and abs denotes an antibiotic sensitive strain. ^aGM denotes the geometric mean of MIC values from all 6 microbial strains in this table.

Denotes fold improvement in the geometric mean of MIC compared to parent peptide ${
m V}_{681}$.

fTherapeutic index = MHC (μg/ml)/geometric mean of MIC (μg/ml). Larger values indicate greater antimicrobial specificity.

 $[^]g$ Denotes fold improvement in the therapeutic index compared to parent peptide ${
m V}_{681}$.

 $^{^{}h}$ In the hemolysis time study, D-V13 K_{D} showed no hemolysis at 500 μ g/ml at 8 hours which would increase the fold improvement in the therapeutic index to 80.2 fold.

^bAntimicrobial and antifungal activities (minimal inhibitory concentration) are from three sets of determinations.

^cS. aureus denotes Staphylococcus aureus, S. epidermidis denotes Staphylococcus epidermidis, B. subtilis denotes Bacillus subtilis, E. faecalis denotes Enterococcus faecalis, C. xerosis denotes Corynebacterium xerosis, C. albicans denotes Candida albicans, wt denotes the wild type strain and methR denotes the methicillin-resistant clinical strain.

^dGM denotes the geometric mean of MIC values from all 6 gram-positive microbial strains in this table. Fold denotes fold improvement in the geometric mean MIC compared to parent

peptide V_{GSI} .

Therapeutic index = MHC (μ g/ml)/geometric mean MIC (μ g/ml). Larger values indicate greater antimicrobial specificity.

^fDenotes the therapeutic index of the peptides against a fungus C. albicans.

 $^{^{}g}$ Denotes fold improvement in the therapeutic index compared to parent peptide V_{681} .

h In the hemolysis time study, D-V13K_D showed no hemolysis at 500 μg/ml at 8 hours which would increase the fold improvement in the therapeutic index to 69.4 fold and 33.4 fold for Gram-positive bacteria and C. albicans, respectively.

blood cells; in contrast, for prokaryotic cells, the peptides lyse cells in a detergent-like mechanism as described in the "car-pet" mechanism.

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Indeed, it is believed that the extent of interaction between peptide and biomembrane is dependent on the composition of 5 lipid bilayer. For example, Liu, et al. (48-50) utilized a polyleucine-based α-helical transmembrane peptide to demonstrate that the peptide reduced the phase transition temperature to a greater extent in phosphatidylethanolamine (PE) bilayers than in phosphatidylcholine (PC) or phosphatidylglycerol (PG) bilayers, indicating a greater disruption of PE organization. The zwitterionic PE is the major lipid component in prokaryotic cell membranes and PC is the major lipid component in eukaryotic cell membranes (51,52). In addition, although PE also exists in eukaryotic membranes, due to 15 the asymmetry in lipid distribution, PE is mainly found in the inner leaflet of the bilayer while PC is mainly found in the outer leaflet of the eukaryotic bilayer. We draw the conclusion that the antimicrobial specificity of the antimicrobial α -helical peptides is a result of the composition differences of the 20 lipid bilayer between eukaryotic and bacterial cells.

In support of this proposal, we have selected two examples from our study. The results for peptide NK_L , the peptide with the highest therapeutic index against Gram-negative bacteria, can be explained using our combined model. For example, if 25 hemolysis of eukaryotic cells requires insertion of the peptide into the hydrophobic core of the membrane, which depends on the composition of the bilayer, and interaction of the non-polar face of the amphipathic α-helix with the hydrophobic lipid environment, it seems reasonable to assume that 30 disruption of the hydrophobic surface with the Lys substitution (NK₁) would both disrupt dimerization of the peptide and its interaction with the hydrophobic lipid. Thus, the peptide is unable to penetrate the hydrophobic core of the membrane and unable to cause hemolysis. On the other hand, if the 35 mechanism for prokaryotic cells allows the interaction of monomeric peptides with the phospholipid headgroups in the interface region, then no insertion into the hydrophobic core of the membrane is required for antimicrobial activity.

The biological activities of the D-enantiomeric peptides 40 illustrated herein are consistent with the proposed model; each enantiomeric peptide pair has the same activities against prokaryotic and eukaryotic cell membranes supporting the prediction that the sole target for these antimicrobial peptides is the cell membrane. This model predicts that hemolysis of eukaryotic cells requires the peptides to be inserted into the hydrophobic core of the membrane, perpendicular to the membrane surface, and interaction of the non-polar face of the amphipathic α -helix with the hydrophobic lipid core of the bilayer. The peptide may thus form transmembrane channels/pores and the hydrophilic surfaces point inward, producing an aqueous pore ("barrel-stave" mechanism). In contrast, antimicrobial activity in prokaryotic cells, while maintaining specificity, requires the peptide to lie at the membrane interface parallel with the membrane surface and interaction of the 55 non-polar face of the amphipathic α-helix with the hydrophobic component of the lipid and interaction of the positively charged residues with the negatively charged head groups of the phospholipid ("carpet" mechanism). What dictates the two different modes of interaction is the difference in lipid 60 composition of prokaryotic and eukaryotic membranes. We refer to this mode of interaction of antimicrobial peptides which combines the above two mechanisms as a "membrane discrimination mechanism".

Using this model, it is easy to understand why peptide NK_L 65 and D- NK_D of the present study are non-hemolytic but at the same time possess excellent antimicrobial activity compared

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to the native sequence V_{681} or D- V_{681} . Thus, the single substitution of Lys for Val at position 13 (NK $_L$ and D-NK $_D$) in the center of the non-polar face disrupts the hydrophobic surface due to the presence of the positive charge, preventing the peptide from penetrating the bilayer as a transmembrane helix in eukaryotic cells. The peptide is then excluded from the bilayer and, hence, is non-hemolytic. In prokaryotic cells, the peptide is also excluded from penetrating the bilayer as a transmembrane helix but this is not required for excellent antimicrobial activity. Instead, the peptide can enter the interface region of the bilayer where disruption of the peptide hydrophobic surface by Lys can be tolerated and antimicrobial activity maintained.

In contrast, the observation that the antimicrobial activity of peptide NL_L (with Leu at the substitution site) was worse than that of NK_L, while its hemolytic activity was stronger (MIC values of 12.7 μ g/ml for NL_L versus 3.1 μ g/ml for NK_L against Gram-negative bacteria; hemolytic activity of 7.8 μ g/ml for NL_L versus no detectable hemolytic activity for NK_L) can also be explained by our combined model. Thus, peptide NL, has a fully accessible non-polar face required for insertion into the bilayer and for interaction with the hydrophobic core of the membrane to form pores/channels ("barrel-stave" mechanism), while the hemolytic activity of peptide NL_L is dramatically stronger than peptide NK_L . On the other hand, due to the stronger tendency of peptide NL_L to be inserted into the hydrophobic core of the membrane than peptide NK_L, peptide NL_L actually interacts less with the water/lipid interface of the bacterial membrane; hence, the antimicrobial activity is 4-fold weaker than the peptide NK, against Gram-negative bacteria. This supports the view that the "carpet" mechanism is essential for strong antimicrobial activity and if there is a preference by the peptide for penetration into the hydrophobic core of the bilayer, the antimicrobial activity will actually decrease.

The studies disclosed herein demonstrate that a high ability of a peptide to self-associate in solution correlates with weak antimicrobial activity and strong hemolytic activity of the peptides. Biological studies further show that strong hemolytic activity of the peptides generally correlates with high hydrophobicity, high amphipathicity and high helicity. In most cases, the D-amino acid substituted peptides possessed an enhanced average antimicrobial activity compared with L-diastereomers. The therapeutic index of V_{681} was improved 90-fold and 23-fold against gram-negative and gram-positive bacteria, respectively. Although the antimicrobial peptides exemplified are the analogs having five amino acid (L, V, A, S, K) substitutions at position 11 or 13 in the 26-residue peptide, V_{681} , it is predicted that other substitutions such as ornithine, arginine, histidine or other positively charged residues at these sites would also improve antimicrobial activity of the peptides. It is further predicted that similar substitutions at position $16 \text{ or } 17 \text{ of } V_{681}$ would yield peptides with enhanced biological activity. Based on the studies disclosed herein, a person of ordinary skill in the art can design antimicrobial peptides with enhanced activities by simply replacing the central hydrophobic or hydrophilic amino acid residue on the nonpolar or the polar face of an amphipathic molecule with a series of selected D-/L-amino acids.

Significant features of two specific antimicrobial peptides generated from this study in structural terms are as follows. In the case of NK_L , a positively-charged residue, lysine, is introduced in the center of the hydrophobic face. This substitution disrupts alpha-helical structure in benign medium, decreases dimerization, decreases toxicity to normal cells as measured by hemolytic activity, enhances antimicrobial activity and provides a 90-fold increase in the therapeutic index compared

with the starting sequence against Gram-negative bacteria (substitution of starting material having Val 13 with a change to Lys 13). The therapeutic index is the ratio of hemolytic activity/antimicrobial activity. This same peptide has a 17-fold increase in the therapeutic index for Gram-positive 5 bacteria.

In the case of NA_D , a D-Ala residue is introduced into the center of the hydrophobic face. This disrupts alpha-helical structure, decreases dimerization, decreases toxicity to normal cells as measured by hemolytic activity, enhances antimicrobial activity and provides a 42-fold increase in the therapeutic index compared to the starting sequence against Gram-negative bacteria (substitution is Val 13 to D-Ala 13). This same peptide has a 23-fold increase in the therapeutic index for Gram-positive bacteria.

Alpha-helical antimicrobial peptides are amphipathic; If the self-association ability of a peptide (forming dimers by interaction of the two non-polar faces of two molecules) is too strong in aqueous media, it could decrease the ability of the peptide monomers to dissociate, pass through the cell wall of 20 microorganisms and penetrate into the biomembranes to kill target cells. It is clearly demonstrated in the studies using the D-enantiomeric peptides that there is a direct correlation of the ability of peptides to dimerize and specificity is generated, that is, disruption of dimerization generates specificity 25 between eukaryotic and prokaryotic cells. From Table 9, the P_A values of peptides derived from their temperature profiling data (FIG. 8) reflect the ability of the amphipathic α -helices to associate/dimerize. Clearly, V_{681} and D- V_{681} , due to their uniform non-polar faces, show the greatest ability to dimerize 30 in aqueous solution and lowest specificity or the strongest ability to lyse human erythrocytes. This is consistent with the view that a peptide with a fully accessible non-polar face tends to form pores/channels in the membranes of eukaryotic cells. In the case of NA_D and D-NA_L, the introduction of 35 D-Ala and L-Ala into all-L- and all-D-amino acid peptides, respectively, disrupts α-helical structure and, thus, lowers dimerization ability relative to V_{681} and D-V $_{681}$ and improves specificity. The introduction of Lys into non-polar position 13 further and improves specificity. Thus, the lack of ability of a peptide to dimerize, as exemplified by its P_A value, is an excellent measure of the peptide's ability to be non-hemolytic concomitant with maintenance of sufficient hydrophobicity of the non-polar face to ensure antimicrobial activity. It is 45 important to note that D-enantiomeric peptides exhibited the same self-association ability as their corresponding L-enantiomers; thus, similar biological activities can be expected. This is further supported by the fact that the hemolytic activity and antimicrobial activity of D-peptides against human 50 red blood cells and microbial cells, respectively, were indeed quantitatively equivalent to those of the L-enantiomers. These results further demonstrate that there is no chiral selectivity by the membrane or other stereoselective interactions in the cytoplasm with respect to the hemolytic and antimicrobial 55 activities.

Because of the different results on the hemolytic activity of peptide NA_D shown in Tables 7 and 8, and Tables 10 and 11 (250 μg/ml after 12 hours versus 31.3 μg/ml after 18 hours, respectively), albeit using the standard microtiter dilution 60 method (see Methods), it became apparent that an investigation of the relationship between hemolysis and time was required. It is noteworthy that there is no universal protocol of determination of hemolytic activity, which makes it difficult to compare data from different sources. For example, some researchers use 4 hours of incubation and take the minimal concentration of peptide to give 100% hemolysis as peptide

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hemolytic activity (28, 56); in contrast, some use 12 hours or longer (e.g., 18 hours used herein) of incubation and take the maximal concentration of peptide to give no hemolysis as peptide hemolytic activity (53, 57). Hence, the hemolysis time study is important to understand the process of erythrocyte lysis. It is clear that the degree of cell lysis is correlated with time, which may be the main reason for the different values of hemolytic activity of NA_D in the two studies. Regardless, the hemolytic activity of each test peptide can readily be appreciated by a skilled artisan by comparing the value of the test peptide with that of the control peptide (V_{681}) within a given study. Hence, we have established a stringent criterion for toxicity, which is no hemolysis at a peptide concentration of 500 µg/ml after 8 hours. We believe that this time study at this very high peptide concentration gives a much more accurate evaluation of hemolytic activity and this method should be established as the gold standard test.

It is important to note that our lead peptides NA_D and NK_D are effective against a diverse group of Pseudomonas aeruginosa clinical isolates. Peptide D-NA₁ exhibited the highest antimicrobial activity against Pseudomonas aeruginosa strains; in contrast, D-NK $_D$ has the best overall therapeutic index due to its lack of hemolytic activity. As mentioned before, Pseudomonas aeruginosa is a family of notorious Gram-negative bacterial strains which are resistant to most of current antibiotics, thus, it is one of the most severe threats to human health (58-60). Only a few antibiotics are effective against Pseudomonas, including fluoroquinolones (61), gentamicin (62) and imipenem (63), and even these antibiotics are not effective against all strains. In the studies disclosed herein, MIC values for *Pseudomonas aeruginosa* and other Gram-negative and Gram-positive bacteria were determined in two different collaborating laboratories; in addition to different media used, the inoculum numbers of cells were also different (see details in the Methods Section), which may explain some variations of MIC values of Pseudomonas aeruginosa strains when comparing the results reported in Table 10 and Table 11.

As shown herein, in general, there is no significant differof NK_L and D- NK_D lowers this dimerization ability even 40 ence in peptide antimicrobial activities against *Pseudomonas* aeruginosa strains, other Gram-negative and Gram-positive bacteria and a fungus between L- and D-enantiomeric peptides, or among peptides with different amino acid substitutions, i.e., V₆₈₁, NA_D and NK_L (Tables 10-12). This observation provides understanding of the mechanism of action of α -helical antimicrobial enantiomeric peptides as follows: there is a dramatic difference in peptide hydrophobicity at position 13 between Val and Lys. The Lys disrupts the continuous non-polar surface due to the positive charge and causes the peptide to locate in the interface region of the microbial membrane. This supports the view that the "carpet" mechanism is essential for strong antimicrobial activity, i.e., for both L- and D-peptide enantiomers, the peptides kill bacteria by a detergent-like mechanism, without penetrating deeply into the hydrophobic core of membrane.

Based on the peptide degradation study, all-D-peptides were totally resistant to enzymatic digestion; hence, this may explain the slightly higher antimicrobial activity of D-peptides than that of their L-enantiomers against Pseudomonas aeruginosa and Gram-positive bacteria. The relatively high susceptibility of L-peptides to trypsin is no doubt due to the presence of multiple lysine residues in sequences, i.e., 6 lysines for V_{681} and NA_D , 7 lysines for NK_L , resulting in the fast degradation of the L-peptides in 30 minutes even at a molar ratio of 20,000:1 (peptide:trypsin).

In summary, by comparing the biophysical and biological properties of L- and D-enantiomeric peptides, we showed that

L- and D-enantiomeric peptide pairs behave the same in selfassociation ability in solution, had the same hemolytic activity against human red blood cells, and exhibited similar antimicrobial activity against *Pseudomonas aeruginosa* strains, and other Gram-negative and Gram-positive bacteria and a 5 fungus. No chiral selectivity was found in the antimicrobial and hemolytic activities of the peptides. Thus, the results disclosed support the "membrane discrimination" model as the mechanism of action for both L- and D-enantiomeric peptides. It is important to note that peptide D-NK_D showed dramatic improvements in therapeutic indices compared to the parent peptide V₆₈₁i.e., 53-fold against Pseudomonas aeruginosa strains, 80-fold against Gram-negative bacteria, 69-fold against Gram-positive bacteria and 33-fold against C. albicans. The proteolytic stability of D-NK $_D$, its broad spec- 15 trum of activity and lack of hemolytic activity demonstrate its clinical potential as a new therapeutic.

Methods for Example 1

Peptide Synthesis and Purification—Syntheses of the peptides Ac-KWKSFLKTFKX $_{D/L}$ AVKTVLHTALKAISS- 20 amide (SEQ ID NO:43) and Ac-KWKSFLKTFKSAX $_{D/L}$ K-TVLHTALKAISS-amide (SEQ ID NO:44), with substitution sites at positions 11 and 13, respectively were carried out by solid-phase peptide synthesis using t-butyloxycarbonyl chemistry and MBHA (4-methylbenzhydrylamine) resin 25 (0.97 mmol/g), as described previously (26). However, it is understood in the art that there are other suitable peptide synthetic devices or that manual peptide synthesis could be carried out to produce the peptides of the present invention. The crude peptides were purified by preparative reversed- 30 phase chromatography (RP-HPLC) using a Zorbax 300 SB- C_8 column (250×9.4 mm I.D.; 6.5 μ m particle size, 300 Åpore size; Agilent Technologies) with a linear AB gradient (0.2% acetonitrile/min) at a flow rate of 2 ml/min, where mobile phase A was 0.1% aqueous TFA in water and B was 0.1% TFA 35 in acetonitrile. The purity of peptides was verified by analytical RP-HPLC as described below. The peptides were further characterized by electrospray mass spectrometry and amino acid analysis.

Analytical RP-HPLC of Peptides—Peptides were analyzed on an Agilent 1100 series liquid chromatograph (Little Falls, Del.). Runs were performed on a Zorbax 300 SB-C $_8$ column (150×2.1 mm I.D.; 5 µm particle size, 300 Å pore size) from Agilent Technologies using linear AB gradient (1% acetonitrile/min) and a flow rate of 0.25 ml/min, where solvent A was 0.05% aqueous TFA, pH 2 and solvent B was 0.05% TFA in acetonitrile. Temperature profiling analyses were performed in 3° C. increments, from 5° C. to 80° C.

Characterization of Helical Structure—The mean residue molar ellipticities of peptides were determined by circular 50 dichroism (CD) spectroscopy, using a Jasco J-720 spectropolarimeter (Jasco, Easton, Md.), at 25° C. under benign conditions (50 mM KH $_2$ PO $_4$ /K $_2$ HPO $_4$ /100 mM KCl, pH 7), as well as in the presence of an α -helix inducing solvent, 2,2,2-trifluoroethanol (TFE) (50 mM KH $_2$ PO $_4$ /K $_2$ HPO $_4$ /100 mM 55 KCl, pH 7 buffer/50% TFE). A 10-fold dilution of a ~500 μ M stock solution of the peptide analogs was loaded into a 0.02 cm fused silica cell and its ellipticity scanned from 190 to 250 nm. The values of molar ellipticities of the peptide analogs at a wavelength of 222 nm were used to estimate the relative α -helicity of the peptides.

CD Temperature Denaturation Study of Peptide V_{681} — The native peptide V_{681} was dissolved in 0.05% aqueous TFA containing 50% TFE, pH 2, loaded into a 0.02 cm fused silica cell and peptide ellipticity scanned from 190 to 250 nm at 65 temperatures of 5, 15, 25, 35, 45, 55, 65 and 80° C. The spectra at different temperatures were used to mimic the

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alteration of peptide conformation during temperature profiling analysis in RP-HPLC. The ratio of the molar ellipticity at a particular temperature (t) relative to that at 5° C. $([\theta]_r-[\theta]_u)/([\theta]_s-[\theta]_u)$ was calculated and plotted against temperature in order to obtain the thermal melting profiles, where $[0]_s$ and $[\theta]_u$ represent the molar ellipticity values for the fully folded and fully unfolded species, respectively. $[\theta]_u$ was determined in the presence of 8M urea with a value of 1500 deg·cm²·dmol⁻¹ to represent a totally random coil state (31). The melting temperature (T_m) was calculated as the temperature at which the α -helix was 50% denatured (($[\theta]_r-[\theta]_u)/([\theta]_s-[\theta]_u)=0.5$) and the values were taken as a measure of α -helix stability.

Determination of Peptide Amphipathicity—Amphipathicity of peptide analogs was determined by the calculation of hydrophobic moment (32) using the software package Jemboss version 1.2.1 (33), modified to include a hydrophobicity scale determined in our laboratory. The hydrophobicity scale used in this study is listed as follows: Trp, 32.31; Phe, 29.11; Leu, 23.42; Ile 21.31; Met, 16.13; Tyr, 15.37; Val, 13.81; Pro, 9.38; Cys, 8.14; Ala, 3.60; Glu, 3.60; Thr, 2.82; Asp, 2.22; Gln, 0.54; Ser, 0.00; Asn, 0.00; Gly, 0.00; Arg, -5.01; His, -7.03; Lys, -7.03 (Hodges, et al. unpublished data). These hydrophobicity coefficients were determined from reversedphase chromatography at pH 2 of a model random coil peptide with single substitution of all 20 naturally occurring amino acids. In this case, the amphipathicity is valid for neutral and acidic pH since V_{681} and analogs do not have Asp and Glu residues in their sequences. We propose that this HPLC-derived scale reflects the relative differences in hydrophilicity/hydrophobicity of the 20 amino acid side-chains more accurately than previously determined scales.

Measurement of Antimicrobial Activity—Minimal inhibitory concentrations (MICS) were determined using a standard microtiter dilution method in LB (Luria-Bertani) no-salt medium (10 g of tryptone and 5 g of yeast extract per liter). Briefly, cells were grown overnight at 37° C. in LB and diluted in the same medium. Serial dilutions of the peptides were added to the microtiter plates in a volume of 100 μl followed by $10 \,\mu l$ of bacteria to give a final inoculum of 5×10^5 colony-forming units/ml. Plates were incubated at 37° C. for 24 hours and MICS determined as the lowest peptide concentration that inhibited growth. Alternatively, minimal inhibitory concentrations were determined using a standard microtiter dilution method in a Mueller-Hinton (MH) medium. Briefly, cells were grown overnight at 37° C. in MH broth and diluted in the same medium. Serial dilutions of the peptides were added to the microtiter plates in a volume of 100 μl followed by 10 μ l of bacteria to give a final inoculum of 1×10⁵ colony-forming units/ml. Plates were incubated at 37° C. for 24 hours and MICS determined as the lowest peptide concentration that inhibited growth. However, for MIC determination of Pseudomonas aeruginosa clinical isolates, brain heart infusion (BH1) medium was used instead of MH broth. In addition, the bacteria were diluted to a final inoculum of 1×10^6 colony-forming units/ml.

Measurement of Hemolytic Activity—Protocol A: Peptide samples were added to 1% human erythrocytes in phosphate buffered saline (0.08M NaCl; 0.043M Na_2PO_4; 0.011 M KH_2PO_4) and reactions were incubated at 37° C. for 12 hours in microtiter plates. Peptide samples were diluted 2 fold in order to determine the concentration that produced no hemolysis. This determination was made by withdrawing aliquots from the hemolysis assays, removing unlysed erythrocytes by centrifugation (800 g) and determining which concentration of peptide failed to cause the release of hemoglobin. Hemoglobin release was determined spectrophoto-

metrically at 562 nm. The hemolytic titer was the highest 2-fold dilution of the peptide that still caused release of hemoglobin from erythrocytes. The control for no release of hemoglobin was a sample of 1% erythrocytes without any peptide added. The results shown in Tables 5-7 were obtained by using Protocol A. Protocol B: Alternatively, peptide samples were added to 1% human erythrocytes in phosphate buffered saline (100 mM NaCl; 80 mM Na₂HPO4; mM NaH₂PO₄, pH 7.4) and reactions were incubated at 37° C. for 18 hours in microtiter plates. Peptide samples were diluted 2 fold in order 10 to determine the concentration that produced no hemolysis. This determination was made by withdrawing aliquots from the hemolysis assays, removing unlysed erythrocytes by centrifugation (800 g) and determining which concentration of peptide failed to cause the release of hemoglobin. Hemoglobin release was determined spectrophotometrically at 570 nm. The hemolytic titer was the highest 2-fold dilution of the peptide that still caused release of hemoglobin from erythrocytes. The control for no release of hemoglobin was a sample of 1% erythrocytes without any peptide added. Since eryth- 20 rocytes were in an isotonic medium, no detectable release (<1% of that released upon complete hemolysis) of hemoglobin was observed from this control during the course of the assay. For the hemolysis time study, hemolytic activity of peptides at concentrations of 8, 16, 32, 64, 125, 250 and 500 25 μg/ml was measured at 0, 1, 2, 4, 8 hours at 37° C. The results shown in Tables 10-12 were obtained by using Protocol B.

Calculation of Therapeutic Index (MHC/MIC Ratio)—it should be noted that both MHC and MIC values were determined by serial 2-fold dilutions. Thus, for individual bacteria and individual peptides the therapeutic index (MHC/MIC, "TI") could vary by as much as 4 fold if the peptide is very active in both hemolytic and antimicrobial activities; if a peptide has poor or no hemolytic activity, the major variation in the therapeutic index (MHC/MIC) comes from the variation in the MIC value (as much as 2-fold).

Proteolytic Stability Assay—Proteolytic stability of the peptides was carried out with trypsin in a molar ratio of 1:20,000 (trypsin:peptide=0.1 µM:2 mM). The buffer used was 50 mM NH₄HCO₃ at pH 7.4 for both peptides and 40 enzyme. The mixtures of peptide and trypsin were incubated at 37° C. Samples were collected at time points of 0, 5 min, 10 min, 20 min, 30 min, 1, 2, 4, 8 hours. Equal volumes of 20% aqueous TFA were added to each sample to stop the reaction and peptide degradation was checked by RP-HPLC. Runs 45 were performed on a Zorbax 300 SB-C₈ column (150×2.1 mm I.D.; 5 µm particle size, 300 Å pore size) from Agilent Technologies at room temperature using a linear AB gradient (1% acetonitrile/min) and a flow rate of 0.25 ml/min, where eluent A was 0.2% aqueous TFA, pH 2 and eluent B was 0.2% 50 TFA in acetonitrile. The change in integrated peak area of the peptides was used to monitor the degree of proteolysis during the time study.

EXAMPLE 2

Peptide Analogs with Varied Position of Substitution

Further peptides of the invention are generated by varying the position of a substitution. By denoting the center position 60 as "i", varied positions of substitutions can be generated while retaining a preferred location on the desired face, e.g. the non-polar face. In the relative context of SEQ ID NO:1, for example, the position for substitution is selected from the group consisting of i, i–4, i–8, i+4, and i+8. See FIG. 7A for 65 examples of specific variants achieved by optional positioning of substitutions. Without wishing to be bound by a par-

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ticular theory, it is hypothesized that a peptide with the substitution at position i of K_L (e.g., in SEQ ID NO:6), in the center position of the non-polar face, can have greater biological activity than a peptide with a substitution at a position further away from the center position. According to such theory, the therapeutic index can decrease in the order of $K_T 13 > K_T 9 > K_T 5$ (here the numeral indicates the position of the amino acid substitution relative to SEQ ID NO:1). Similarly, the therapeutic index can decrease in the order of $A_D 13 > A_D 9 > A_D 5$ (here $A_D 13$ corresponds to SEQ ID NO:9). Regardless of such theory, such peptides with varied positions of a substitution can have activity and be useful in compositions and methods of the invention. Successful substitutions at position 13 can also work near the center of the hydrophobic face at position 9, 12, 16 and 17 (see helical net diagram in FIG. 1).

In order to evaluate the biological activities of the peptide analogs with varied position of substitution, we used peptide NK_L as a framework to systematically alter the peptide hydrophobicity by replacing alanine residues with hydrophobic leucine residues on the non-polar face of the helix. Peptide sequences are shown in Table 13.

TABLE 13

	Sequences of 1	NK, analog peptides.
Num- ber	Denotation	Peptide sequences a
1	$NK_{L (SEQ ID NO: 6)}$	Ac-KWKSFLKTFKSAKKTVLHTA LKAISS-amide
2	A23L _(SEQ ID NO: 45)	Ac-KWKSFLKTFKSAKKTVLHTA LK $_{ m L}$ ISS-amide
3	A12L _(SEQ ID NO: 46)	Ac-KWKSFLKTFKS <u>I</u> KKTVLHTA LKAISS-amide
4	A20L _(SEQ ID NO: 47)	Ac-KWKSFLKTFKSAKKTVLHT $_{f L}$ LKAISS-amide
5	A12L/ A23L _(SEQ ID NO: 48)	Ac-KWKSFLKTFKS $_{f L}$ KKTVLHTA LK $_{f L}$ ISS-amide
6	A12L/ A20L _(SEQ ID NO: 49)	Ac-KWKSFLKTFKS $_{f L}$ KKTVLHT $_{f L}$ LKAISS-amide
7	A12L/A20L/ A23L _(SEQ ID NO: 50)	

⁴Peptide sequences are shown using the one-letter code for amino acid residues; Ac- denotes N^4 -acetyl and -amide denotes C^4 -amide; in sequences, bold letters denote the substituting amino acids at the non-polar face of NK_L (see FIG. 1 for details).

FIG. 12 shows the leucine-substituted peptide analogs represented as helical nets. Since there are three alanine residues on the non-polar face of peptide V13K₁, three single Leusubstituted analogs were synthesized with an Ala→Leu substitution at positions 12, 20 and 23; two double Leu-substituted peptides (A12L/A23L and A12L/A20L) and a triple Leu-substituted peptide (A12L/A20L/A23L) were also made to increase hydrophobicity. In FIG. 12, the $i\rightarrow i+3$ and $i\rightarrow i+4$ hydrophobic interactions are shown as black bars. It is clear that, for single Leu-substituted analogs, the number of hydrophobic interactions created by these substitutions in the three peptides is in the order of A20L>A12L>A23L (9, 8 and 6 hydrophobic interactions, respectively); for peptides with double leucine substitutions, A12L/A20L has more hydrophobic interactions than A12L/A23L (11 versus 8 hydrophobic interactions); peptide A12L/A20L/A23L presented the highest number of i→i+3 and i→i+4 hydrophobic interac-

tions (12 hydrophobic interactions) among all the peptide analogs as expected. In this study, peptides were simply named after the positions of the leucine substitutions.

For the Leu-substituted peptides, the hemolytic activity against human erythrocytes was determined as the maximal 5 peptide concentration that produces no hemolysis after 18 hours of incubation at 37° C. Antimicrobial activity of the peptides was determined on a diverse group of Pseudomonas aeruginosa clinical isolates. Pseudomonas is a family of strains that has become more and more resistant to various 10 traditional antibiotics since 1990s and it is also known to produce proteolytic enzymes which make it even less susceptible to antimicrobial peptides (58-60). Pseudomonas aeruginosa strains used in this study exhibited a wide range of ciprofloxacin susceptibility (a 64-fold difference) (Table 14). 15 The most resistant Pseudomonas aeruginosa strain was CP204, a clinical isolate from cystic fibrosis patients, which was, in contrast, the most susceptible strain to our antimicrobial peptides in this study. This indicates that there is no correlation between antibiotic resistance and the ability of 20 antimicrobial peptides to be effective. The geometric mean MIC values from six *Pseudomonas* strains in Table 14 were calculated to provide an overall evaluation of antimicrobial activity of the peptides with different hydrophobicities.

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action for antimicrobial peptides whose sole target is the biomembrane. We believe that the mechanism depends upon the compositional difference in the lipids between prokaryotic and eukaryotic membranes. It is well-known that eukaryotic cell membranes are in contrast to prokaryotic membranes generally characterized by zwitterionic phospholipids, a relatively large amount of cholesterol and sphigomyelin, and the absence of a high, inside-negative transmembrane potential presented in prokaryotic membranes (51-52, 66-67). Hence, if the peptides form pores/channels in the hydrophobic core of the eukaryotic bilayer, they would cause the hemolysis of erythrocytes; in contrast, for prokaryotic cells the peptides lyse cells in a detergent-like mechanism as described in the carpet mechanism (46).

The observation that there is a correlation between peptide hydrophobicity and hemolytic activity can be explained by the "membrane discrimination" mechanism. Peptides with higher hydrophobicity will penetrate deeper into the hydrophobic core of red blood cell membrane (67), causing stronger hemolysis by forming pores or channels, i.e., A12L/A23L (peptide 5) and A12L/A20L (peptide 6) exhibited stronger hemolytic activity than single Leu-substituted peptides, and A12L/A20L/A23L (peptide 7) showed the strongest hemolytic activity in this study (Table 14). For peptide anti-

TABLE 14

		Biologica	al activiti	es of NK	. analog	g peptide	es.				
			Antimicrobial activity								
	Peptide	Hemolytic ac	tivity			М	IC (μg/m	l) ^b			_
Numbe	r Denotation	MHC (μg/ml)	Fold^a	PAO 1	WR 5	PAK	PA 14	M 2	CP 204	GM^c	Fold^d
1	NK_L	250.0	1.0	31.3	250.0	125.0	62.5	31.3	7.8	49.6	1.0
2	A23L	62.5	4.0	31.3	62.5	62.5	31.3	31.3	15.6	35.1	1.4
3	A12L	31.3	8.0	15.6	62.5	31.3	31.3	15.6	15.6	24.8	2.0
4	A20L	31.3	8.0	7.8	31.3	31.3	15.6	15.6	7.8	15.6	3.2
5	A12L/A23L	15.6	16.0	15.6	250.0	15.6	62.5	31.3	7.8	31.2	1.6
6	A12L/A20L	8.0	32.0	62.5	125.0	62.5	125.0	62.5	62.5	78.7	0.6
7	A12L/A20L/A23L	4.0	62.5	500.0	500.0	500.0	500.0	500.0	500.0	500.0	0.1
	Ciprofloxacine			4	1	4	8	8	64		

^adenotes the fold increase in hemolytic activity compared to the parent peptide NK_L. Hemolytic activity (MHC, the maximal peptide concentration showing no hemolysis after 18 hours) was determined on human red blood cells (hRBC).
^bMIC is minimal inhibitory concentration against 6 Pseudomonas aeruginosa clinical isolates.

The influence of peptide hydrophobicity on biological activities is shown in FIG. 13. It was generally accepted that increasing the hydrophobicity of the non-polar face of the amphipathic α -helical peptides would increase both 50 hemolytic and antimicrobial activities (5-8). In the present study, it is clear that peptide hydrophobicity correlated with peptide hemolytic activity in that the more hydrophobic the peptide the stronger the hemolytic activity to erythrocytes, which is consistent with previous results (53, 64-66). The 55 striking result from our antibacterial assay was that hydrophobicity has two effects on peptide antimicrobial activity: at a relative low level of hydrophobicity, increasing peptide hydrophobicity caused an improvement in antimicrobial activity until an optimum hydrophobicity was reached; in 60 contrast, peptide antimicrobial activity was weakened dramatically with further increases of hydrophobicity beyond the optimum, even resulting in the loss of antimicrobial activity of peptide A12L/A20L/A23L (peptide 7) in this study (FIG. 12).

The results of these studies are consistent with our proposed model of "membrane discrimination" mechanism of microbial activity, since the insertion of the molecules into the hydrophobic core is not necessary to lyse bacterial cells during the antibacterial action, peptides only lie at the interface parallel with the membrane allowing their hydrophobic surface to interact with the hydrophobic component of the lipid, and the positive charge residues to interact with the negatively charged head groups of the phospholipids (46,47). Thus, it is reasonable to assume that increasing peptide hydrophobicity to a certain extent will help peptide molecules to reach the interface from aqueous environment and improve antimicrobial activity. In this study, the improvement of antimicrobial activity from peptide NK_L (peptide 1) to peptide A20L (peptide 4) can represent such an advantage of increasing hydrophobicity. In contrast, further increases in hydrophobicity will cause the stronger peptide dimerization in solution which in turn results in the monomer-dimer equilibrium favoring the dimer conformation. Peptide dimers are in their folded α -helical conformation and would be inhibited from passing through the cell wall to reach the target membranes. Hence the antimicrobial activities of peptides A12L/A23L (peptide 5) and A12L/A20L (peptide 6) become weaker with increas-

^cdenotes the geometric mean of the MIC values for the 6 Pseudomonas aeruginosa clinical isolates.

ddenotes the fold improvement in antimicrobial activity (geometric mean data) compared to the parent peptide NK_I

eRelative susceptibility of these Pseudomonas strains to ciprofloxacin with the number 1 denoting the most susceptible strain.

ing hydrophobicity compared to the single Leu-substituted analogs. We believe that there is a threshold of hydrophobicity controlling peptide antimicrobial activity, that is, one may adjust peptide hydrophobicity to obtain the optimal antimicrobial activity. For the extreme example of the triple-Leusubstituted analog, A12L/A20L/A23L (peptide 7), the loss of antimicrobial activity may be explained as due to its very strong dimerization ability in aqueous environments. Hence, the peptide exists mainly as a dimer in solution and it would not pass through the bacterial cell wall. In contrast, there is no polysaccharide-based cell wall in eukaryotic cells, thus, A12L/A20L/A23L (peptide 7) caused severe hemolysis against human red blood cells where the hydrophobicity of the bilayer causes rapid dissociation of dimers to monomers and entry into the bilayer to form channels/pores.

EXAMPLE 3

Peptide Analogs with Varied Nature of Charge Substitution

Further peptides of the invention are generated by varying the nature of the charged residue selected for the substitution. In the relative context of SEQ ID NO:1, for example, the position for substitution is established as position 13. The 25 amino acid selected for substitution is preferably a charged amino acid and is in particular an amino acid with a net positive charge. Particular examples of charged residues at position 13 are Lys, Arg, Orn, His, diamino butyric acid and diamino propionic acid. We note that Orn has a delta/ δ -amino group instead of an epsilon/E-amino group in Lys, i.e., the side-chain is shorter by one carbon atom; diaminobutanoic acid is one carbon shorter than Orn; i.e., it has a gamma/ γ -amino group; diaminopropionic acid is two carbons shorter than Orn, i.e., it has a beta/ β -amino group.

See FIG. 7A for a specific variant peptide achieved by optional selection of charged residues for use in substitutions. A peptide with a charged residue in the center of the non-polar face can be active. Without wishing to be bound by a particular theory, it is hypothesized that the activity of a peptide with such a centrally positioned positively charged residue can be modulated depending on the positively charged residue, although there may be difficulty in predicting the precise effect upon an activity parameter such as the therapeutic index as described herein.

EXAMPLE 4

Peptide Analogs with Multiple Substitutions

Further peptides of the invention are generated by using multiple substitutions relative to a reference sequence such as SEQ ID NO:1. In a preferred embodiment, the multiple substitutions are a double substitution. In the relative context of SEQ ID NO:1, for example, the following peptides are made 55 with double substitutions: a) peptide with substitution combination of L6 to A_D6 and L21 to A_D21 ; and b) peptide with substitution combination of L6 to K_L6 and L21 to K_L21 . See FIG. 7A for specific variant peptides achieved by optional multiple substitutions.

Without wishing to be bound by a particular theory, it is hypothesized that the activity of a peptide with multiple substitutions (e.g. two substitutions) not in the center position can still be effective. For a particular peptide generated by multiple substitutions, such multiple substitutions can be at least as effective as a single substitution in the center of the non-polar face. Alternatively, a given multiple substitution such as

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the specific double substitutions shown may not be as effective as the single substitutions described herein due to the removal of two Leu residues instead of one Val residue. A decrease in hydrophobicity can optionally result in a decrease in the therapeutic index. In addition, the double D-Ala substitutions may be more disruptive of the helical structure; such disruption can also yield a decrease in the therapeutic index. Analogous results can be achieved for the double L-Lys substitutions.

EXAMPLE 5

Truncated Peptide Analogs

Further peptides of the invention are generated by truncation of a reference peptide such as SEQ ID NO:1 or a peptide of the invention such as any of SEQ ID NOS: 2 to 25 (except for those peptides that are exactly equivalent to SEQ ID NO:1 such as SEQ ID NO: 3 and 15), or other peptide of the invention as described herein. In a truncated peptide analog, for example, truncation of the N-terminal residue Lys1 or C-terminal residues Ser25 and Ser26 do not substantially affect the biological properties such as activity of the truncated peptide. See FIG. 7A for specific variant peptides achieved by optional truncations in the relative context of peptide NK_L (SEQ ID NO:6). Without wishing to be bound by a particular theory, it is hypothesized that truncating the N-terminal residue Lys1 or C-terminal residues Ser25 and Ser26 does not substantially affect the biological properties. It is believed, however, that truncation of Lys1 and Trp2 can substantially decrease the therapeutic index due to removal of the large hydrophobe, Trp. Similarly, truncation of Ser26, Ser25 and Ile24 can substantially decrease the therapeutic index due to removal of the large hydrophobe, Ile.

As further alternatives to truncated analog peptides corresponding to SEQ ID 34, 35, 36, and 37, there is a substitution of K13V (the K at position 13 is replaced by V) for each of those respective peptides.

EXAMPLE 6

Shuffled Peptide Analogs

Further peptides of the invention are generated by shuffling various residues. For residues on the non-polar face, shuffling is performed to generate various non-polar face analogs. It is believed that the overall hydrophobicity of the non-polar face is an important factor in biological activity. The hydrophobic residues on the non-polar face are shuffled wherein a resulting peptide is biologically active. As shown in FIG. 7A and FIG. 7B, the circled residues in the sequence are the shuffled residues on the non-polar face.

For residues on the polar face, shuffling is performed to generate various polar face analogs. It is believed that the polar face is an important factor in maintaining the amphipathicity of the molecule and significant properties such as biological activity. The polar residues on the polar face are shuffled wherein a resulting peptide is biologically active. As shown in FIG. 7A and FIG. 7C, the circled residues in the sequence are the shuffled residues on the polar face.

Shuffling of the non-polar face residues or polar face residues does not substantially change the amphipathicity of the molecule, or the overall hydrophobicity of the non-polar face or the overall hydrophilicity of the polar face. Though a particular sequence may be a preferred or the best analog, other analogs can nonetheless have useful biological properties.

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In particular peptides, shuffling the 11 residues on the non-polar face is performed, excluding the substitution site (for example so as to preserve β -Lys as in NK_L), to any combination of sequences that contain 1 W (one tryptophan residue), 2 F, 3 L, 3 A, 1 V and 1 I residue. Similarly, shuffling the 14 residues on the polar face to any combination of sequences that contains 6 K, 4 S, 3 T, and 1 H residue is performed.

Peptides are generated having a range of overall hydrophobicity of the non-polar face. The hydrophobicity of the non-polar face can be calculated using a sum of the hydrophobicity coefficients listed in the manuscript. For example, a particular hydrophobicity range is of NK_L or NA_D±the value of a Leu side-chain. Using our scale, the hydrophobicity of 15 the non-polar face of NK_L sums up the values for W2, F5, L6, F9, A12, K13, V16, L17, A20, L21, A23, I24 getting a value of 199.7. See Table 15.

TABLE 15

Hydrophobicity coefficients.										
	Item Coefficient									
	Trp 2	32.31								
	Phe 5	29.11								
	Leu 6	23.42								
	Phe 9	29.11								
	Ala 12	3.60								
	Lys 13	-7.03								
	Val 16	13.81								
	Leu 17	23.42								
	Ala 20	3.60								
	Leu 21	23.42								
	Ala 23	3.60								
	Ile 24	21.31								
	SUM	199.7 ± 23.42								

Different scales can give different values. For peptides herein, there is significance in the sum of the residues in the hydrophobic surface, using our scale, where the surface hydrophobicity range that generates the desired biological activity is from about 176 to about 224.

The sum of the hydrophobicity coefficients for the polar face should be the value for NK_L peptide±the value of a Lys residue.

TABLE 16

	Coefficient values.								
Item Coefficient									
	K1	-7.03							
	K3	-7.03							
	S4	0.00							
	K7	-7.03							
	Т6	+2.82							
	K10	-7.03							
	S11	0.00							
	K14	-7.03							
	T15	+2.82							
	H18	-7.03							
	T19	+2.82							
	K22	-7.03							
	S25	0.00							
	S26	0.00							
	SUM	-40.75 ± 7.03							

Using our scale, the hydrophobicity of the polar face of NK_L sums up the values K1, K3, S4, K7, T6, K10, S11, K14,

T15, H18, T19, K22, S25 and S26. The range of surface hydrophilicity that generates the desired biological activity is from about -33 to about -48.

EXAMPLE 7

Peptide Analogs with Similar Single Hydrophobicity Substitutions

Further peptides of the invention are generated by making single substitutions of amino acid residues with relatively similar hydrophobicity. Single hydrophobicity substitutions with side-chains of similar hydrophobicity are generated and have biological activity. For example, possible substitutions for each residue in the non-polar face are listed below in Table 17 in the context of peptides NK_L and NA_D (SEQ ID NOS:6 and 9, respectively).

TABLE 17

Residues for single substitutions.						
idue in NK_L or NA_D	Substituting residues					
	Ile, Val, norleucine, norvaline					
	Leu, Val, norleucine, norvaline					
	Leu, Ile, norleucine, norvaline					
	Leu, Ile, Val, norleucine, norvaline					
	Phe, Leu, Ile, Val, norleucine, norvaline					
	$\frac{\mathrm{Resio}}{\mathrm{Idue in NK}_L} \text{ or NA}_D$					

EXAMPLE 8

Peptide Analogs of NA_D and D-NA_L

Further compositions and methods are provided where the hemolytic activity of NA_D or $\mathrm{D}\text{-}\mathrm{NA}_L$ is further decreased by decreasing the overall hydrophobicity of the non-polar face. See Kondejewski, L. H., et al. 2002, J. Biol. Chem. 277:67-74. For example, V16 is substituted to A16; or L17 to A17; or both V16, L17 to A16, A17. A decreased hydrophobicity can decrease hemolytic activity as shown for substitutions herein at position 13. The hydrophobicity can decrease approximately in the order $\mathrm{NL}_L > \mathrm{NV}_Q > \mathrm{NA}_L > \mathrm{NG} > \mathrm{NS}_L > \mathrm{NK}_L$ which correlates with the weakening of hemolytic activity (µg/ml) where NL_L (7.8), NV_Q (15.6), NA_L (31.2), NG (125), NS_A (125) and NK_L (no measurable activity). It is recognized that there can be a threshold of hydrophobicity which when excessively decreased can result in peptides where the biological property of antimicrobial activity is substantially reduced.

STATEMENTS REGARDING INCORPORATION BY REFERENCE AND VARIATIONS

All references throughout this application, for example patent documents including issued or granted patents or equivalents; patent application publications; and non-patent literature documents or other source material; are hereby incorporated by reference herein in their entireties, as though individually incorporated by reference, to the extent each reference is at least partially not inconsistent with the disclosure in this application (for example, a reference that is partially inconsistent is incorporated by reference except for the partially inconsistent portion of the reference).

Any appendix or appendices hereto are incorporated by 65 reference as part of the specification and/or drawings.

Where the terms "comprise", "comprises", "comprised", or "comprising" are used herein, they are to be interpreted as

specifying the presence of the stated features, integers, steps, or components referred to, but not to preclude the presence or addition of one or more other feature, integer, step, component, or group thereof.

The invention has been described with reference to various 5 specific and preferred embodiments and techniques. However, it should be understood that many variations and modifications may be made while remaining within the true spirit and scope of the invention. It will be apparent to one of ordinary skill in the art that compositions, methods, devices, device elements, materials, procedures and techniques other than those specifically described herein can be applied to the practice of the invention as broadly disclosed herein without resort to undue experimentation. All art-known functional equivalents of compositions, methods, devices, device elements, materials, procedures and techniques described herein are intended to be encompassed by this invention. It is not intended, however, for any claim herein to specifically encompass any precise embodiment existing and legally 20 28. Zhang, L., Benz, R., and Hancock, R. E. (1999) Biochemqualifying in the relevant jurisdiction as prior art for novelty; a claim purportedly encompassing such an embodiment is intended to be of scope so as to just exclude any such precise embodiment.

Whenever a range is disclosed, all subranges and indi- 25 vidual values are intended to be encompassed. This invention is not to be limited by the embodiments disclosed, including any shown in the drawings or exemplified in the specification, which are given by way of example or illustration and not of limitation.

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The Sequence Listing is incorporated by reference herein.

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<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 15
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 16
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 16
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Val Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 17
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEOUENCE: 17
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Lys Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
            20
```

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<210> SEQ ID NO 18
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(11)
<223> OTHER INFORMATION: L at position 11 is D-Leu.
<400> SEQUENCE: 18
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Leu Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 19
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(11)
<223> OTHER INFORMATION: A at position 11 is D-Ala.
<400> SEOUENCE: 19
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ala Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
            20
<210> SEQ ID NO 20
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(11)
<223> OTHER INFORMATION: S at position 11 is D-Ser.
<400> SEQUENCE: 20
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 21
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11) . . (11)
<223> OTHER INFORMATION: V at position 11 is D-Val.
<400> SEQUENCE: 21
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Val Ala Val Lys Thr Val
                            10
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
            20
```

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<210> SEQ ID NO 22
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(11)
<223> OTHER INFORMATION: K at position 11 is D-Lys.
<400> SEQUENCE: 22
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Lys Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 23
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 23
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Gly Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
           2.0
<210> SEQ ID NO 24
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (1)..(26)
<223> OTHER INFORMATION: All amino acids are in D-conformation.
<400> SEQUENCE: 24
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 25
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (1)..(26)
<223> OTHER INFORMATION: Each amino acid except A at position 13 is in
     D-conformation.
<400> SEQUENCE: 25
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Ala Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
            20
```

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<211> LENGTH: 18
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 26
Glu Leu Glu Lys Gly Gly Leu Glu Gly Glu Lys Gly Gly Lys Glu Leu
                                    10
Glu Lys
<210> SEQ ID NO 27
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 27
Lys Trp Lys Ser Phe Leu Lys Thr Lys Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
           20
<210> SEQ ID NO 28
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 28
Lys Trp Lys Ser Lys Leu Lys Thr Phe Lys Ser Ala Val Lys Thr Val
                                 10
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
           20
<210> SEQ ID NO 29
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (9)..(9)
<223> OTHER INFORMATION: A at position 9 is D-Ala.
<400> SEQUENCE: 29
Lys Trp Lys Ser Phe Leu Lys Thr Ala Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
          20
<210> SEQ ID NO 30
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (5)..(5)
<223> OTHER INFORMATION: A at position 5 is D-Ala.
<400> SEQUENCE: 30
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Lys Trp Lys Ser Ala Leu Lys Thr Phe Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 31
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 31
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Arg Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 32
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (1)..(26)
<223> OTHER INFORMATION: A at positions 6 and 21 is D-Ala.
<400> SEQUENCE: 32
Lys Trp Lys Ser Phe Ala Lys Thr Phe Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Ala Lys Ala Ile Ser Ser
           20
<210> SEQ ID NO 33
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 33
Lys Trp Lys Ser Phe Lys Lys Thr Phe Lys Ser Ala Val Lys Thr Val
Leu His Thr Ala Lys Lys Ala Ile Ser Ser
<210> SEQ ID NO 34
<211> LENGTH: 25
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 34
Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val Leu
                       10
His Thr Ala Leu Lys Ala Ile Ser Ser
           20
<210> SEQ ID NO 35
<211> LENGTH: 24
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
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<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 35
Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val Leu His
                                   10
Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 36
<211> LENGTH: 24
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 36
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile
<210> SEQ ID NO 37
<211> LENGTH: 23
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 37
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val
                                    10
Leu His Thr Ala Leu Lys Ala
           2.0
<210> SEQ ID NO 38
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 38
Lys Ile Lys Ser Ala Leu Lys Thr Leu Lys Ser Phe Lys Lys Thr Ala
Ala His Thr Leu Phe Lys Val Trp Ser Ser
<210> SEQ ID NO 39
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 39
Ser Trp Ser Lys Phe Leu Lys Lys Phe Thr Lys Ala Lys Ser His Val
Leu Thr Thr Ala Leu Ser Ala Ile Lys Lys
<210> SEQ ID NO 40
<211> LENGTH: 26
<212> TYPE: PRT
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<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(13)
<223> OTHER INFORMATION: X at positions 11 and 13 is Gly, L- or D-Leu,
      L- or D-Val, L- or D-Ser, L- or D-Ala, L- or D-Lys.
<400> SEQUENCE: 40
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Xaa Ala Xaa Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 41
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 41
Lys His Ala Val Ile Lys Trp Ser Ile Lys Ser Ser Val Lys Phe Lys
Ile Ser Thr Ala Phe Lys Ala Thr Thr Ile
            20
<210> SEQ ID NO 42
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEOUENCE: 42
His Trp Ser Lys Leu Leu Lys Ser Phe Thr Lys Ala Leu Lys Lys Phe
Ala Lys Ala Ile Thr Ser Val Val Ser Thr
           20
<210> SEQ ID NO 43
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(11)
<223> OTHER INFORMATION: X at position 11 is any amino acid in L- or
      D-conformation.
<400> SEQUENCE: 43
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Xaa Ala Val Lys Thr Val
                                  10
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 44
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
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<222> LOCATION: (13)..(13)
<223> OTHER INFORMATION: X at position 13 is any amino acid in L- or
     D-conformation.
<400> SEQUENCE: 44
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Xaa Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
        20
<210> SEQ ID NO 45
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 45
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val
                     10
Leu His Thr Ala Leu Lys Leu Ile Ser Ser
          20
<210> SEQ ID NO 46
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 46
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Leu Lys Lys Thr Val
Leu His Thr Ala Leu Lys Ala Ile Ser Ser
           20
<210> SEQ ID NO 47
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 47
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Lys Thr Val
Leu His Thr Leu Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 48
<211> LENGTH: 26
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 48
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Leu Lys Lys Thr Val
        5
                        10
Leu His Thr Ala Leu Lys Leu Ile Ser Ser
           20
<210> SEQ ID NO 49
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<211> LENGTH: 26

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<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<400> SEOUENCE: 49
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Leu Lys Lys Thr Val
Leu His Thr Leu Leu Lys Ala Ile Ser Ser
<210> SEQ ID NO 50
<211> LENGTH: 26
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 50
Lys Trp Lys Ser Phe Leu Lys Thr Phe Lys Ser Leu Lys Lys Thr Val
Leu His Thr Leu Leu Lys Leu Ile Ser Ser
<210> SEQ ID NO 51
<211> LENGTH: 22
<212> TYPE: PRT
<213 > ORGANISM: Artificial sequence
<220> FEATURE:
<223 > OTHER INFORMATION: Synthetic peptide
<400> SEQUENCE: 51
Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Leu Lys Thr Val Leu
His Thr Ala Leu Lys Ala
            20
<210> SEQ ID NO 52
<211> LENGTH: 22
<212> TYPE: PRT
<213> ORGANISM: Artificial sequence
<220> FEATURE:
<223> OTHER INFORMATION: Synthetic peptide
<220> FEATURE:
<221> NAME/KEY: MISC_FEATURE
<222> LOCATION: (11)..(11)
<223> OTHER INFORMATION: K at position 11 is D-lysine.
<400> SEQUENCE: 52
Lys Ser Phe Leu Lys Thr Phe Lys Ser Ala Lys Leu Lys Thr Val Leu
His Thr Ala Leu Lys Ala
            20
```

What is claimed is:

- 1. A method of increasing antimicrobial activity or decreasing hemolytic activity of an antimicrobial peptide comprising
 - producing a modified antimicrobial peptide by replacing at least one hydrophobic amino acid residue in a hydrophobic face of the antimicrobial peptide of SEQ ID NO:1 with at least one hydrophilic amino acid residue,
 - wherein the modified antimicrobial peptide produced is 65 helical in a hydrophobic environment, and disordered in aqueous media.
- 2. The method of claim 1, wherein the at least one hydrophilic amino acid residue is an amino acid residue having a negatively-charged side chain.

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- 3. The method of claim 1, wherein the at least one hydrophilic amino acid residue is an amino acid residue having a positively-charged side chain.
- **4**. The method of claim **3**, wherein the at least one hydrophilic amino acid residue is a lysine amino acid residue.
- 5. The method of claim 1, wherein the at least one hydrophilic amino acid residue is selected from the group consist-

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ing of D-lysine and D-alanine, and wherein all other amino acids residues in the modified antimicrobial peptide are in the L-enantiomeric form.

- 6. The method of claim 1, wherein the modified antimicrobial peptide produced is selected from the group consisting of 5 SEQ ID NO:24 and SEQ ID NO:25.
- 7. The method of claim 1, wherein the modified antimicrobial peptide produced is selected from the group consisting of SEQ ID NO:6 and SEQ ID NO:9.
- **8**. The method of claim **1**, wherein the modified antimicrobial peptide produced is selected from the group consisting of SEQ ID NO:10 and SEQ ID NO:11.
- **9**. The method of claim **1**, wherein the producing step further comprises truncating one residue from an end of the antimicrobial peptide of SEQ ID NO:1.
- 10. The method of claim 1, wherein the producing step further comprises truncating two residues from an end of the antimicrobial peptide of SEQ ID NO:1.

* * * * *